

CORAL REEFS UNDER GLOBAL CHANGE

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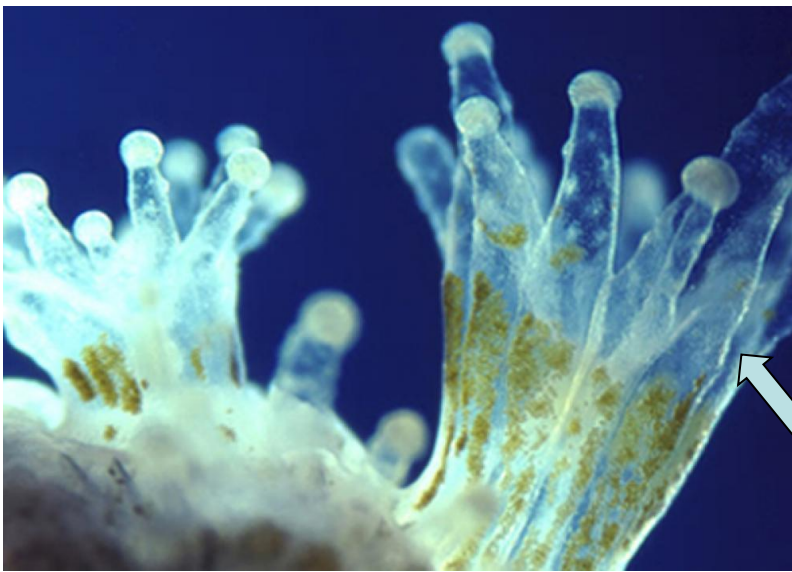




Coral reefs

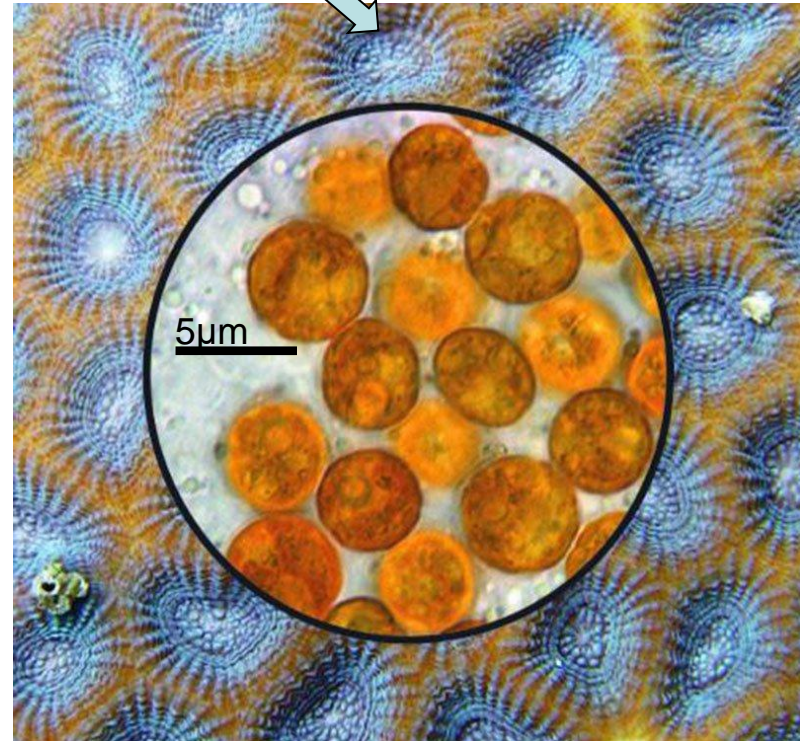
- The most diverse and complex ecosystem in the ocean with thousands of species
- Highly productive system. Rate of gross photosynthesis is 50-100 time higher than in the next by open ocean
- Multiple symbiotic relations from the cell level to the ecosystem level
- An important component of the global carbon cycle, despite its small area (0.2%)
- Economically important (tourism, fisheries etc)





The corals belong to the Coelenterates (Medusa, Anemonies). Most of them are colonial built of thousands of **polyps** having stinging tentacles and myriads of symbiotic algae (10^6 cm^{-2}). They build unique CaCO_3 skeletons that build the coral reefs.

One polyp with symbiotic algae



Light Enhanced Calcification

(the common dogma)

CO₂ is taken by the algae for photosynthesis and this promotes calcification
(and vice versa: calcification enhances photosynthesis)

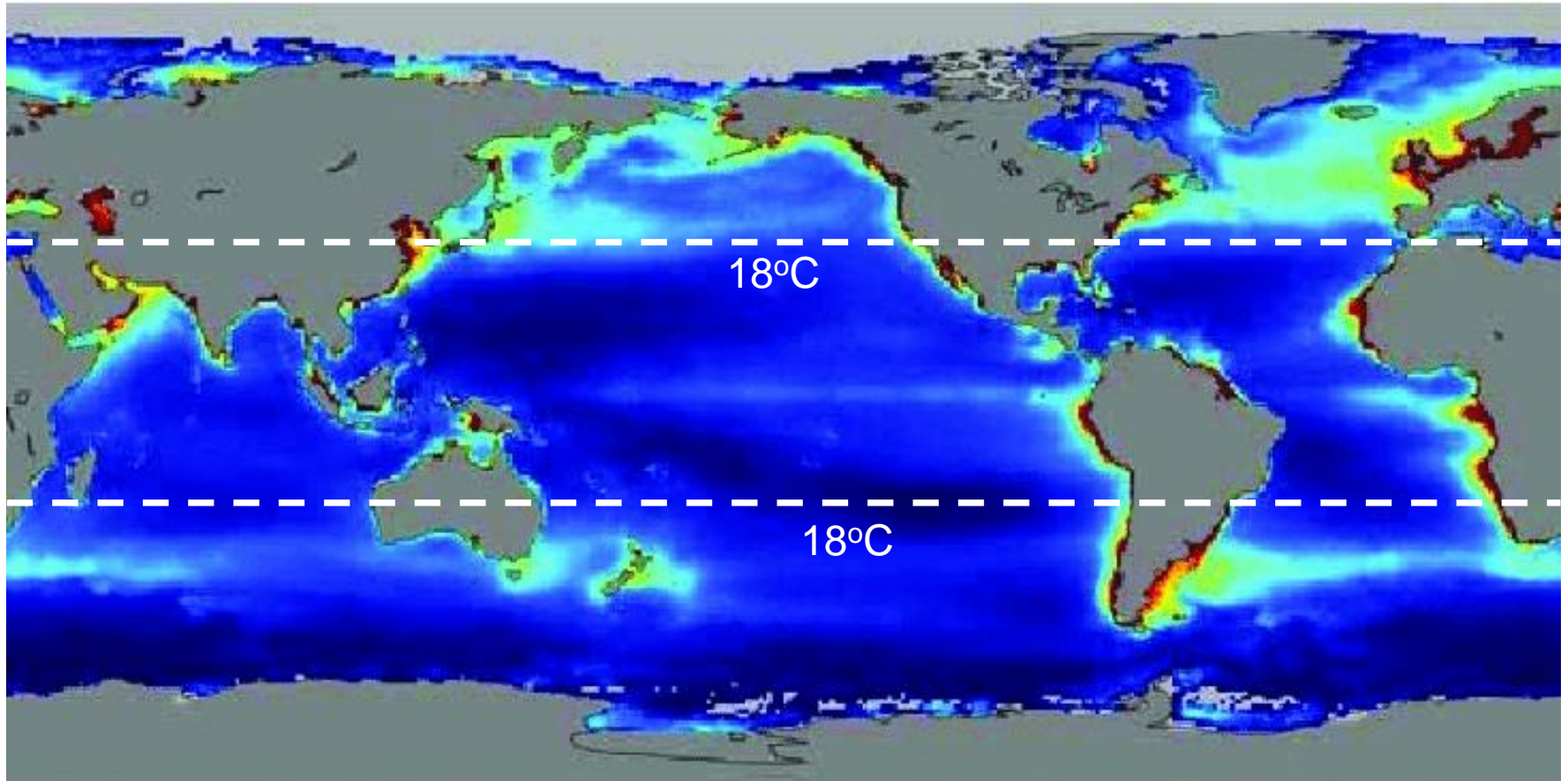
Calcification



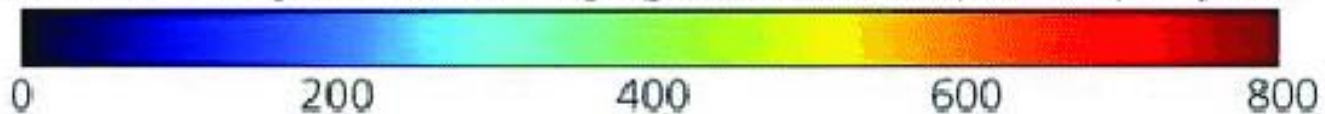
Photosynthesis

After Goreau TF, 1959

Coral reefs are found in the oligotrophic parts of the ocean – the blue “desert”



Net Primary Productivity (grams Carbon per m² per year)



The nutrient paradox of coral reefs 1

- The ecologist Odum was the first to point this paradox: “OASIS IN THE DESERT OCEAN”
- Coral reefs thrive in subtropical waters where nutrient concentrations are very low, and primary productivity is $0.1-0.2 \text{ gC m}^{-2} \text{ day}^{-1}$.
- Primary productivity within coral reefs is $10-15 \text{ gC m}^{-2} \text{ day}^{-1}$, 100 times more than in the surrounding water.
- The traditional ecological explanation: Highly efficient nutrient recycling within the coral reef ecosystem
- But coral reefs are open systems constantly flushed by ocean currents tides and waves

A coral reefs is an “Oasis in the desert ocean” (Odum 1971)

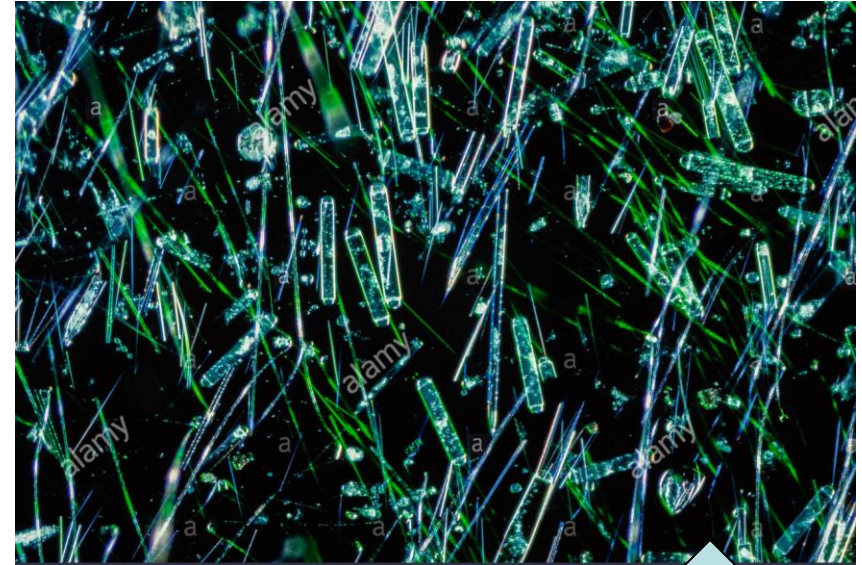
What is the cause for their ecological success?



Solving the nutrient paradox of coral reefs

- We proposed that a coral reef is a huge filtering system that extract **plankton** from the oligotrophic water.
- This mega-filter is built by the corals (and other calcifiers). It stands on a solid ground and creates a wave resistant structure.
- The higher is the flow and water exchange with the open sea more plankton is supplied to the ecosystem.
- The nutrients are coming from the **particulate phase** and dissolved nutrients are released to the open sea

PLANKTON



alamy stock photo

W716MW
www.alamy.com

Zooplankton (Copepods etc.)

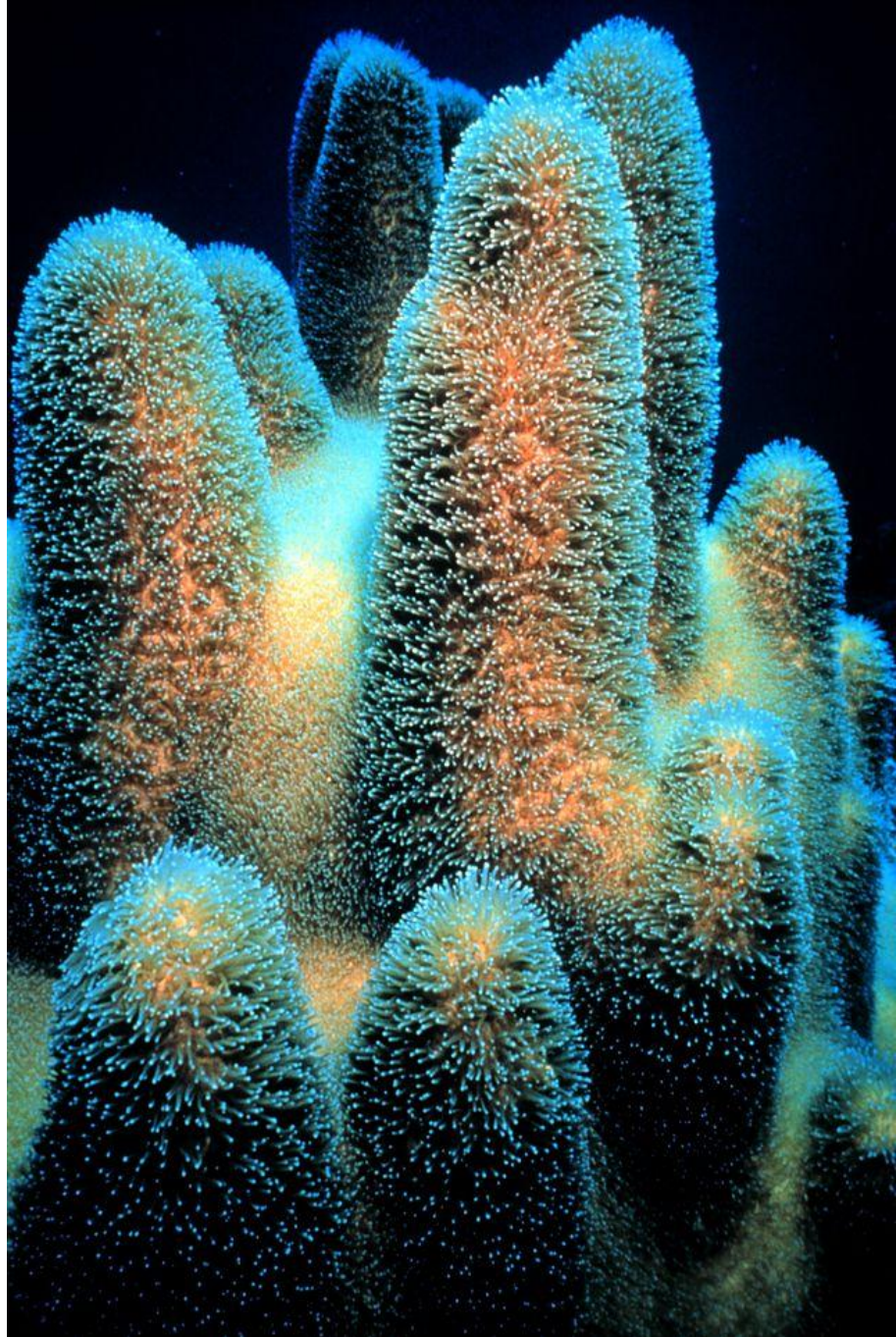
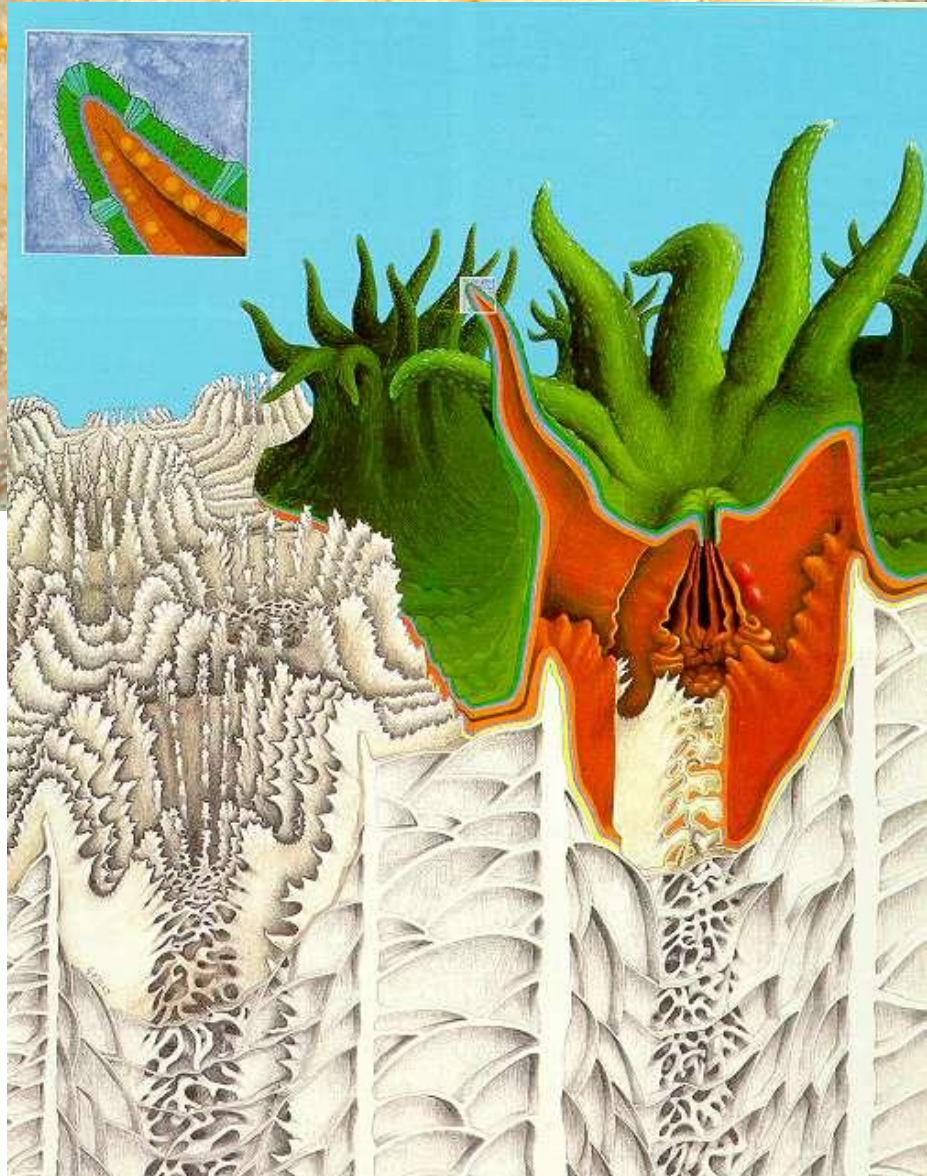
Phytoplankton (Diatoms)

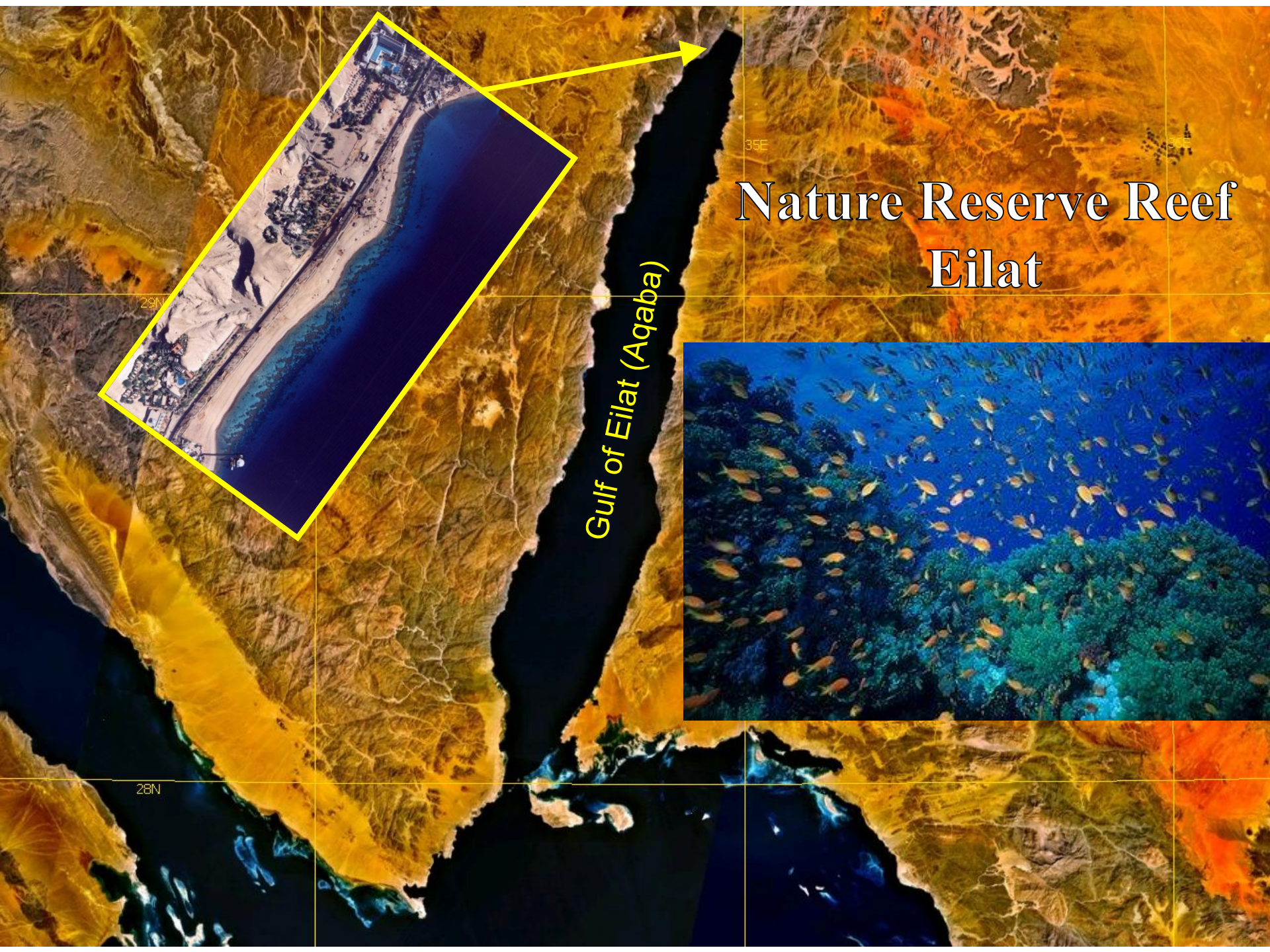


The coral reef is a huge filtering system for oceanic plankton



Photo: C. Verone





Nature Reserve Reef Eilat

Gulf of Eilat (Aqaba)

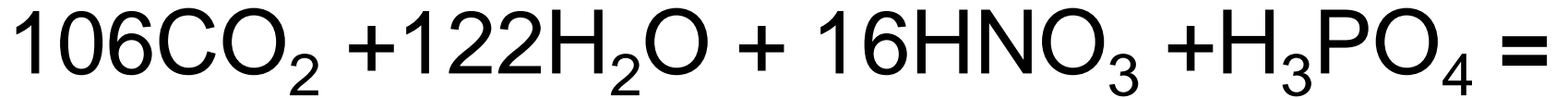


Other plankton feeders in the reef



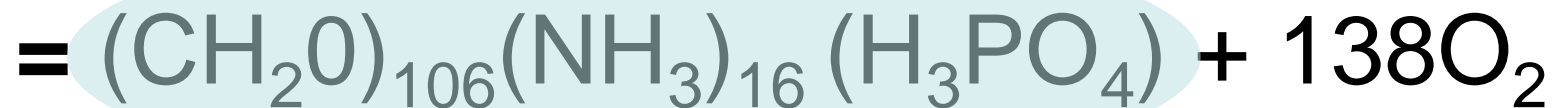
The Modified Redfield Equation

→
PHOTOSYNTHESIS



←
RESPIRATION

Geochemical organic matter



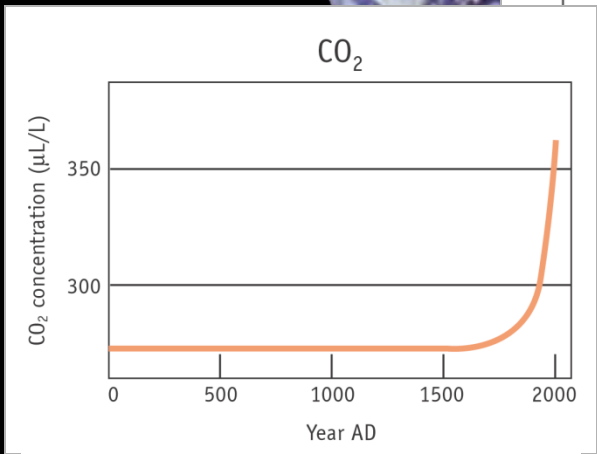
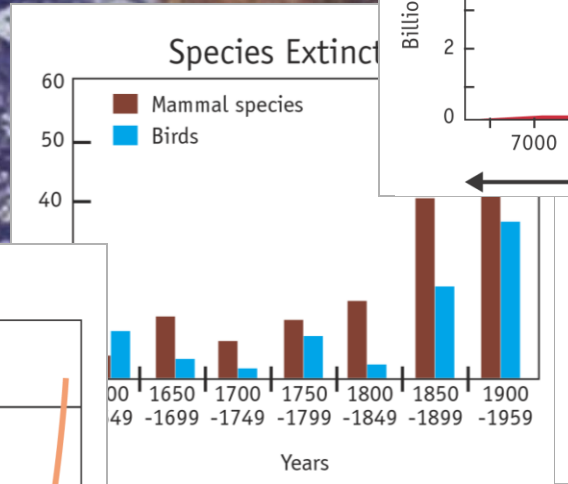
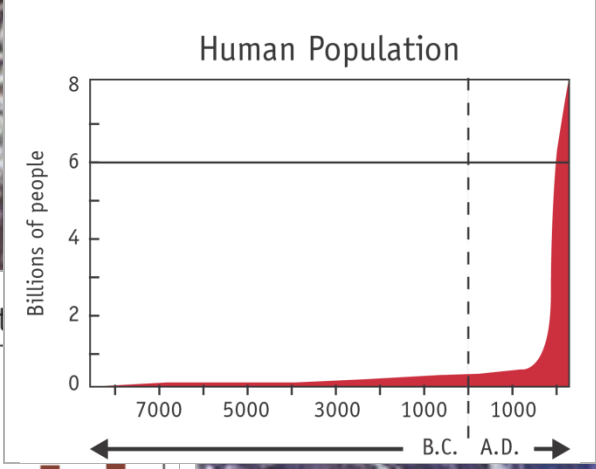
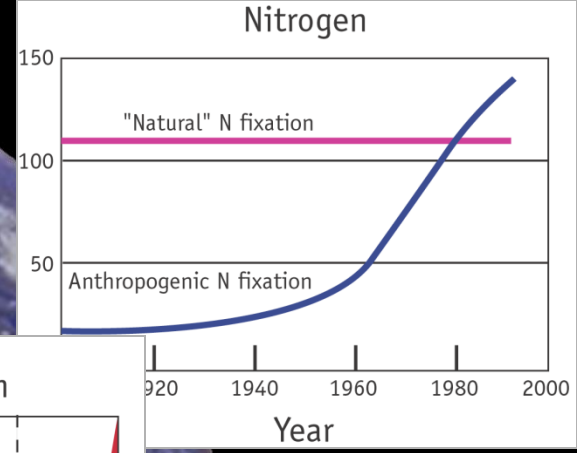
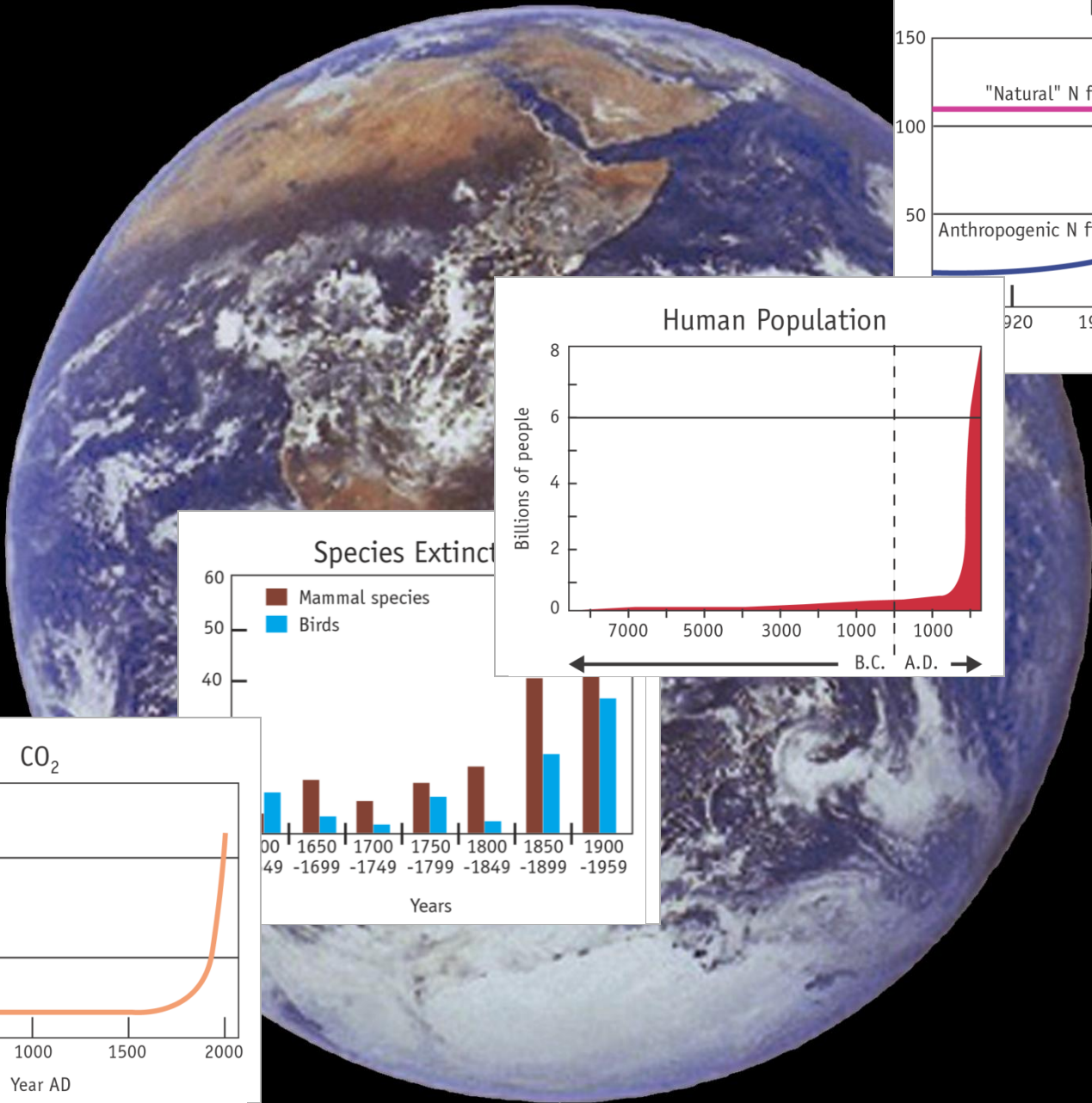


The key for coral reef ecosystem function is the rigid framework produced by the corals which serves as a mega-filtration system

Photo: I. Greenberg

Summary: Coral reefs, obtain their nutrients from the oceanic plankton

- Their success depends on **their solid reef framework** that allows massive bio-filtration of plankton from the intense oceanic water flows
- This requires intensive calcification by the corals and other framework builders to compensate for the bio-erosion and physical damage of storm
- *What will happen to this framework under ocean acidification?*

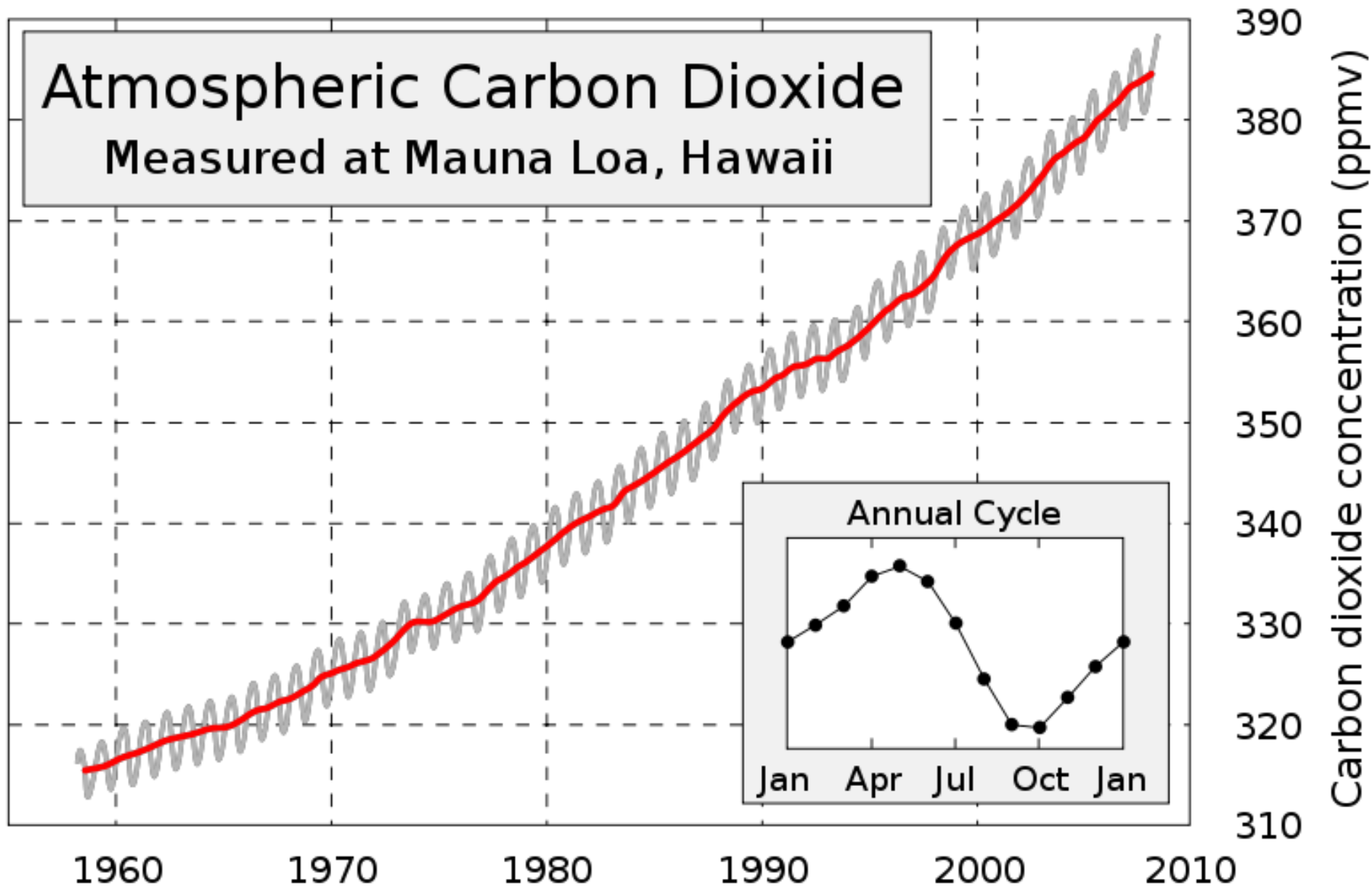


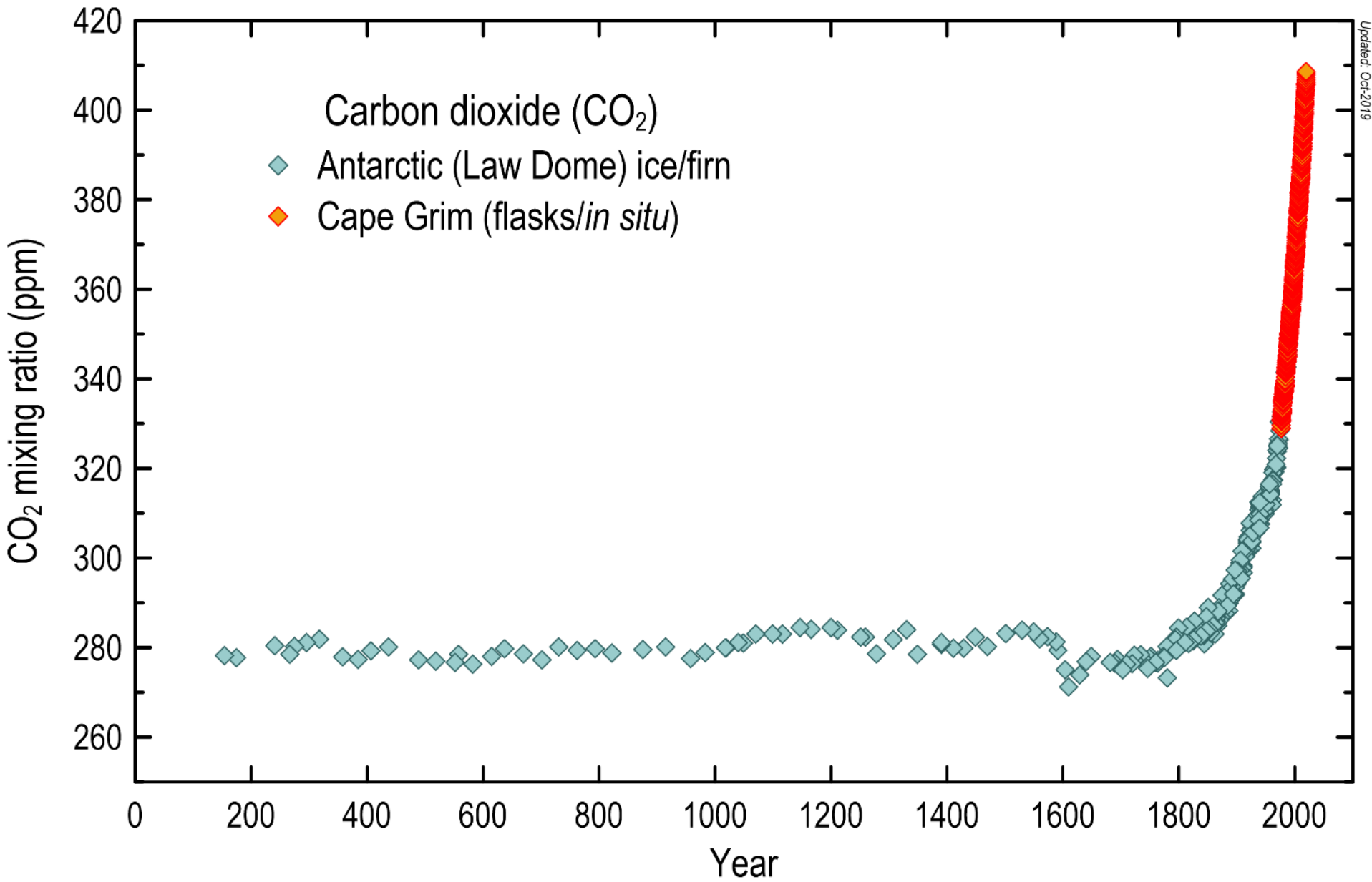
Source Moore

Influence of global change on coral reefs

- Eutrophication from sewage and mari-culture cause algal blooms that smother the corals
- CO₂ increase lead to ocean acidification and reduced calcification of corals and others
- Global warming is causing coral bleaching and mass mortality of corals

Atmospheric Carbon Dioxide Measured at Mauna Loa, Hawaii





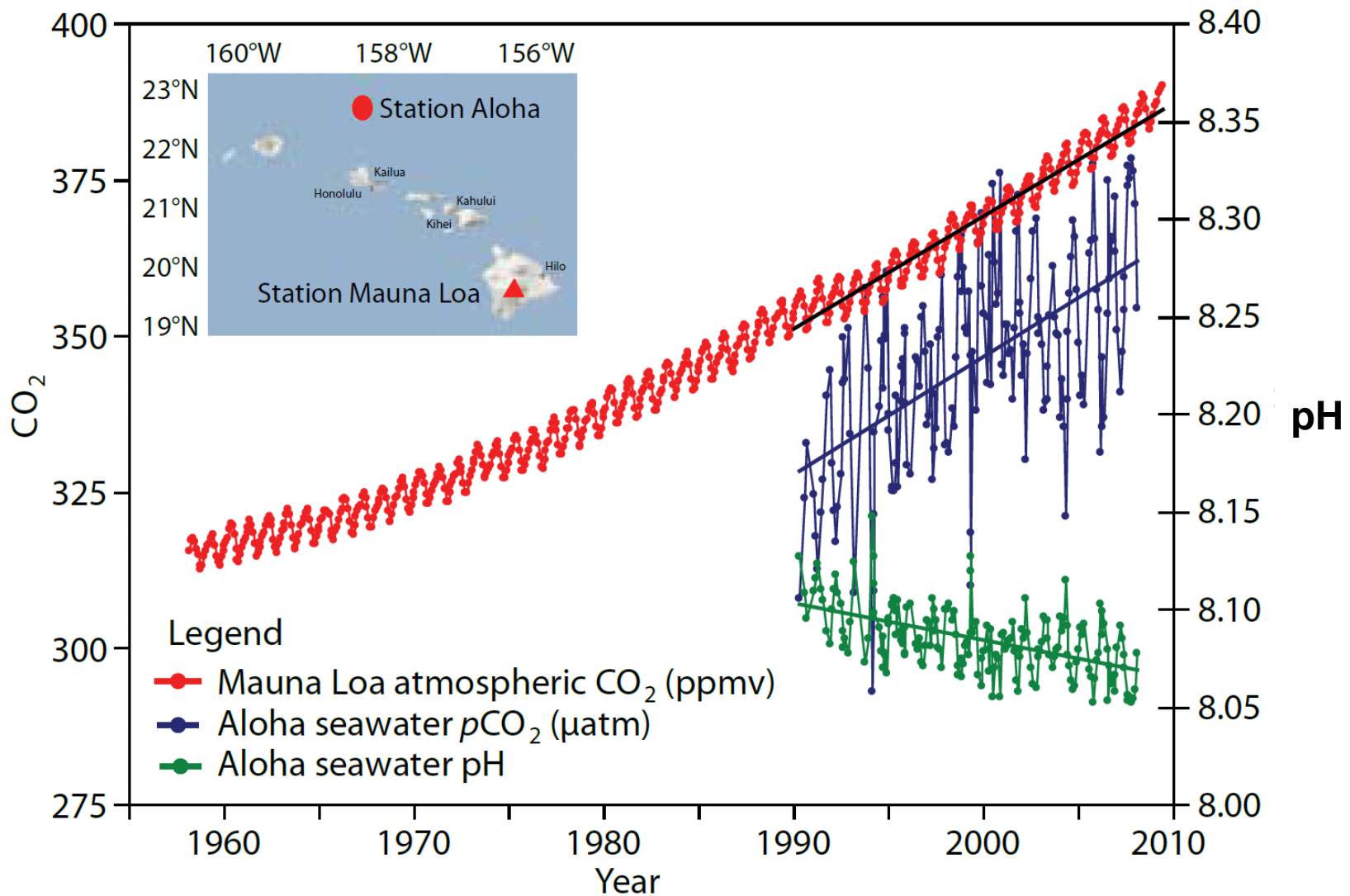
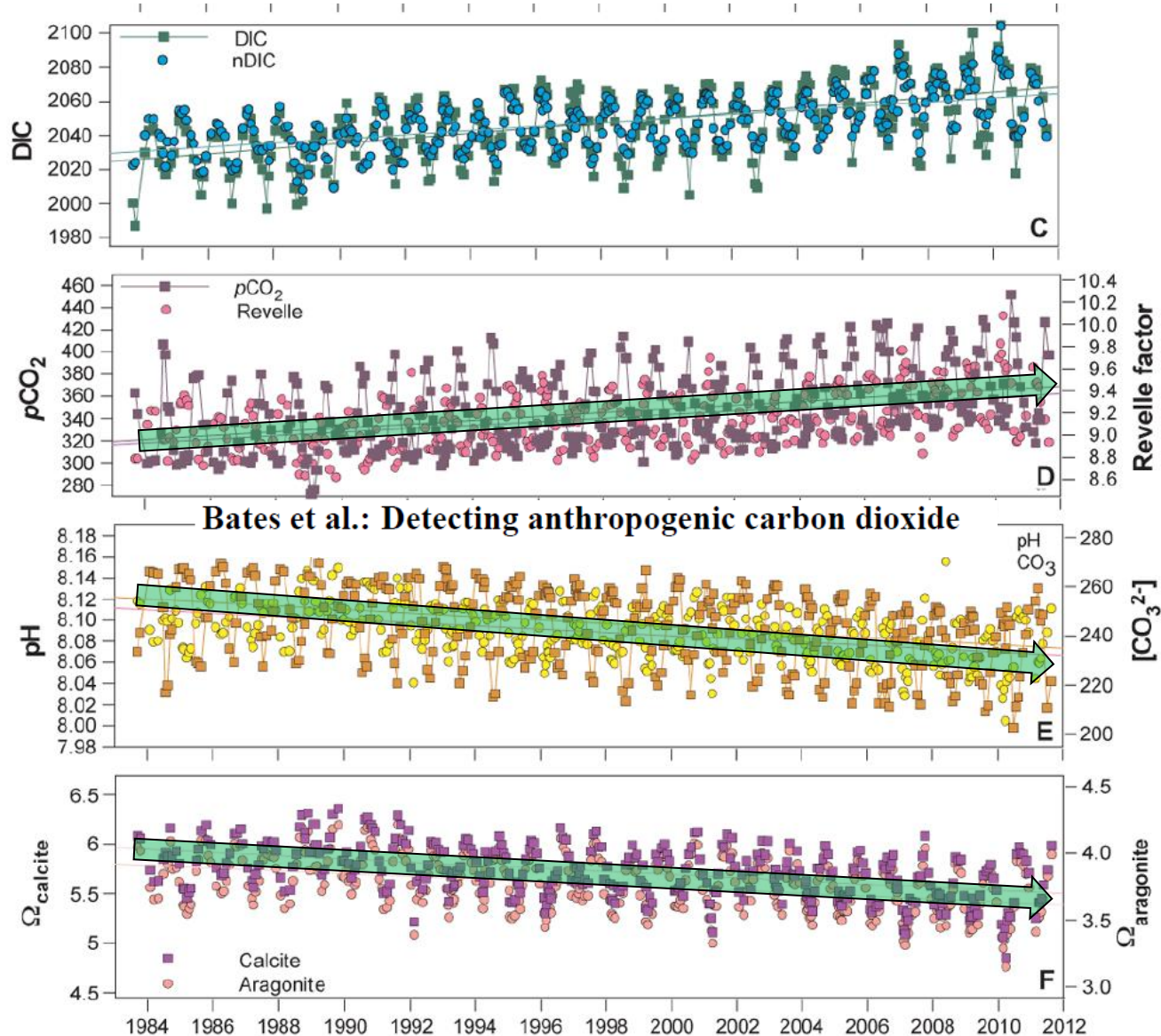


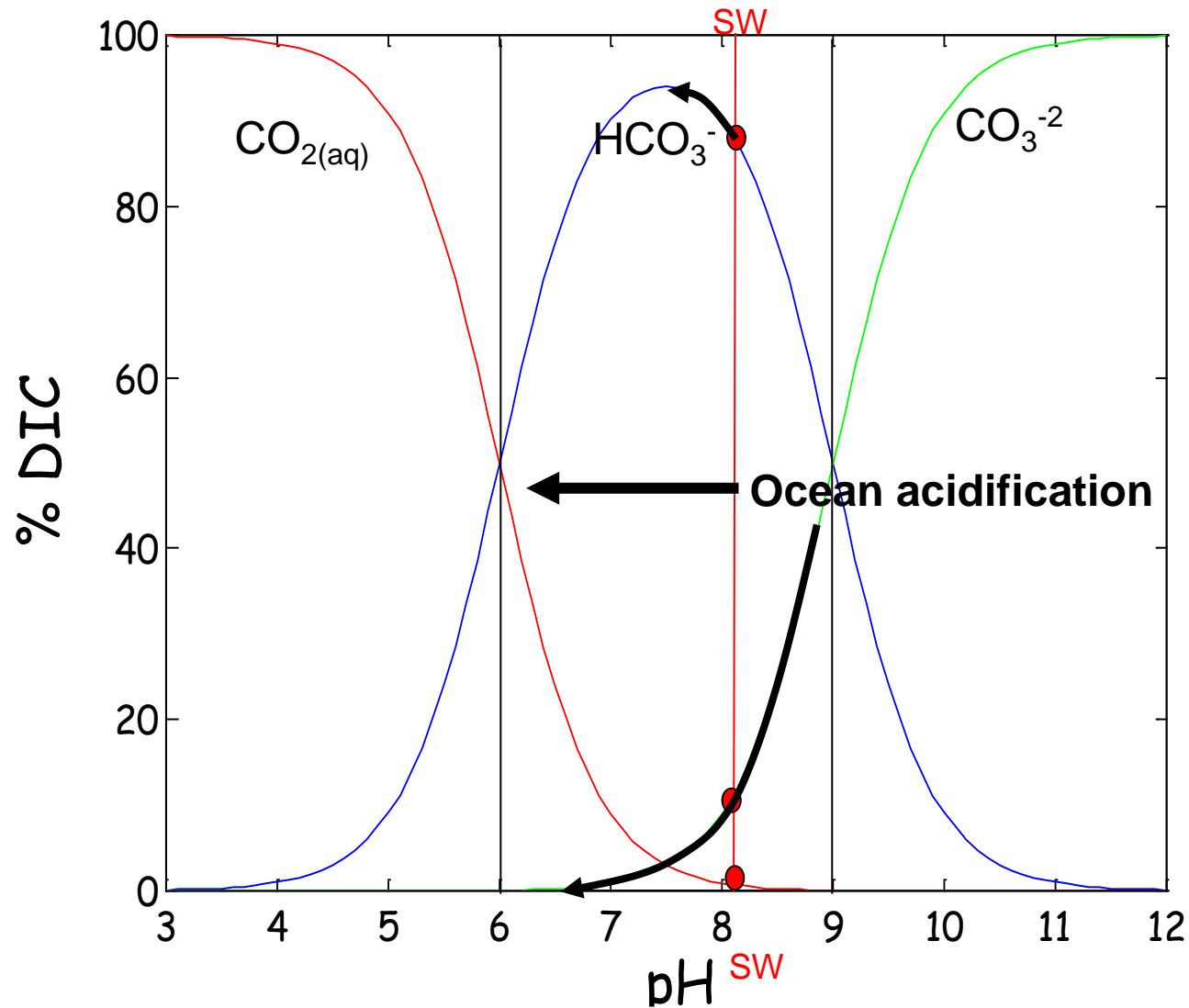
Figure 1: Time series of atmospheric CO₂ at Mauna Loa (in parts per million volume, ppmv; red), surface ocean pCO₂ (μatm; blue) and surface ocean pH (green) at Ocean Station ALOHA in the subtropical North Pacific Ocean.

Doney et al 2009, Oceanography

BATS Bermuda Atlantic time series: 1984 - 2012



Equilibrium of the carbonate system in the ocean



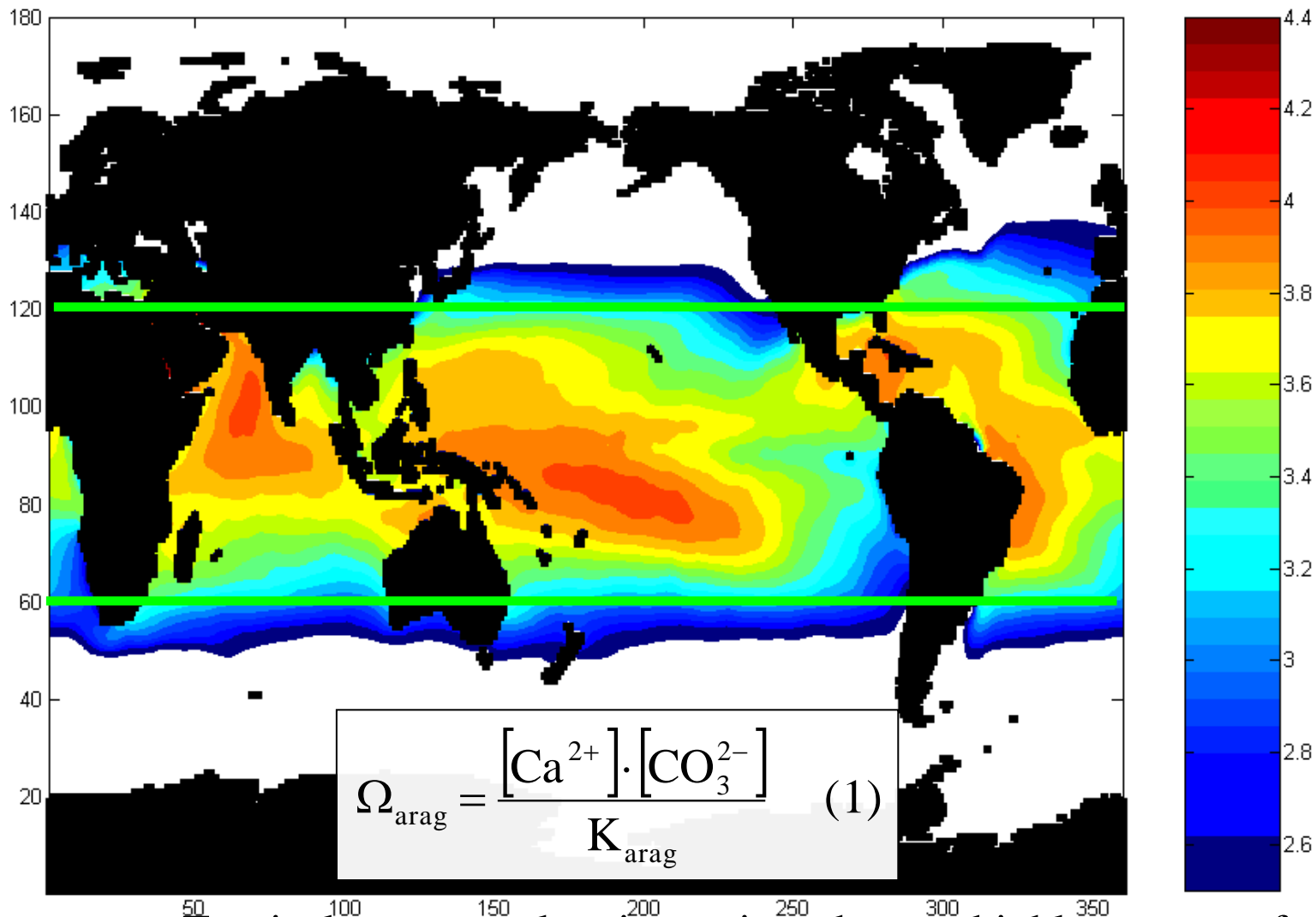
The dependence of coral calcification on CaCO_3 (aragonite) saturation

$$\Omega_{arag} = \frac{[Ca^{2+}] \cdot [CO_3^{2-}]}{K_{arag}} \quad (1)$$

Ca in the ocean is high (10-11 mM) and ~ constant. Therefore the CO_3^{2-} ion is controlling Ω_{arag}

Despite supersaturation ($\Omega_{arag} = 3-4$) reefs start to dissolve when Ω_{arag} is below 3

מידת הרוויה של מי ים ביחס לגיר שמשקיעים אלמוגים: מה שמסביר את תפוצתם הגיאוגרפית – חרב פיפיות!



Levitus 1998

Tropical water are low in nutrients but are highly saturates for the mineral aragonite (CaCO_3) $\Omega_{\text{arag}} > 3.5$

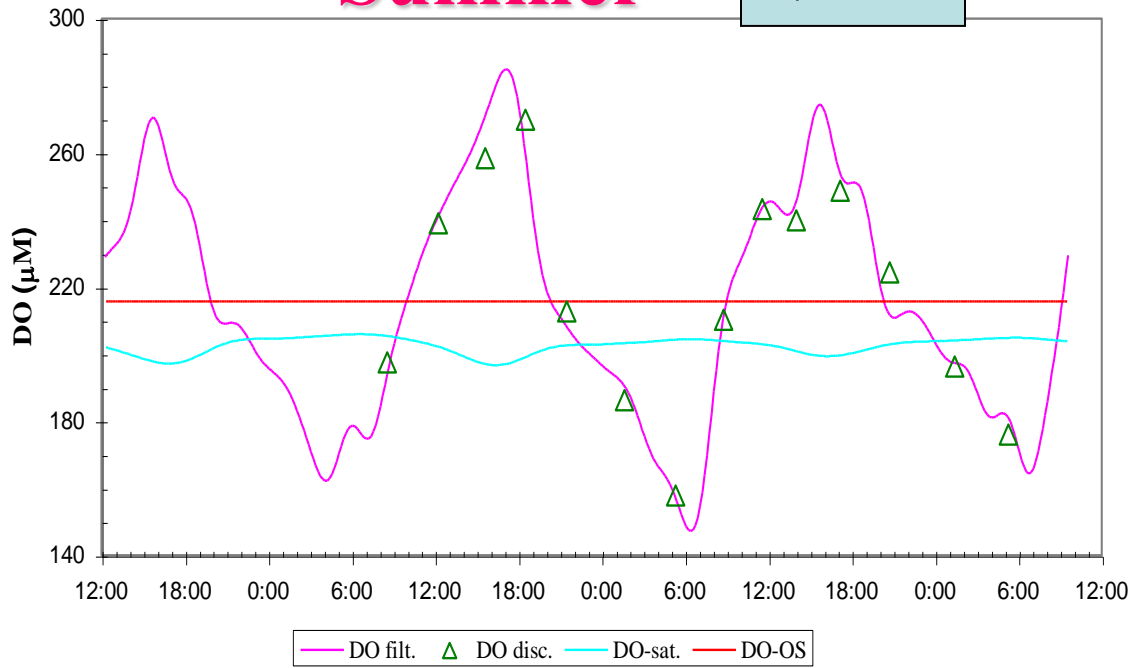
Community metabolism in the Eilat reef

Oriya Barzel MSc



Summer

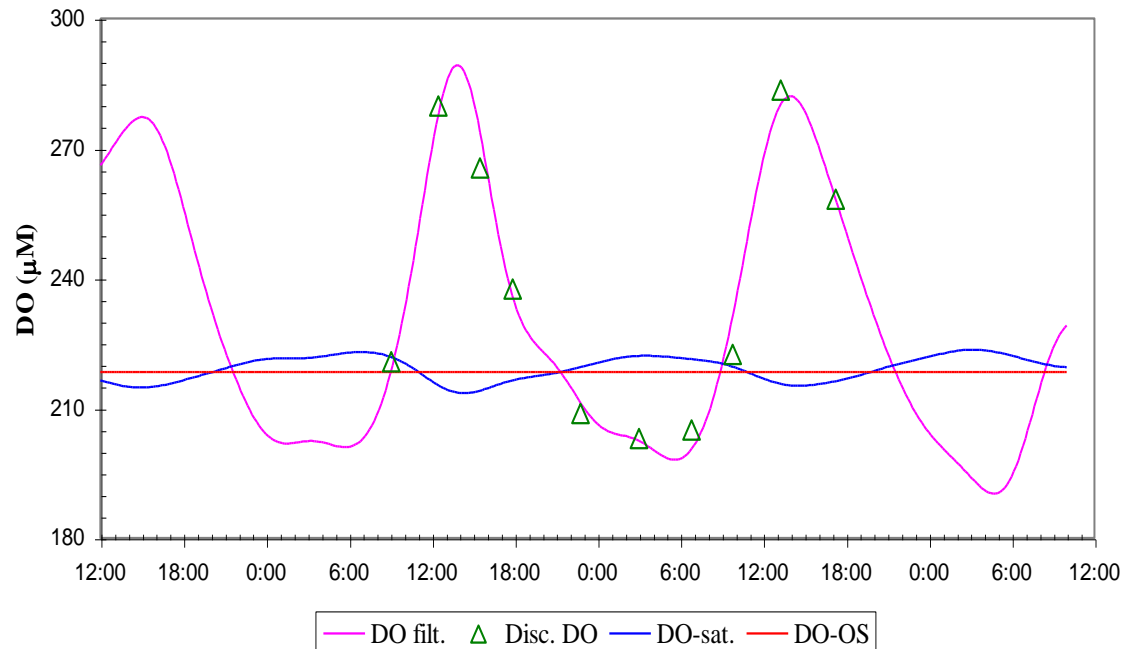
$P/R \approx 1$

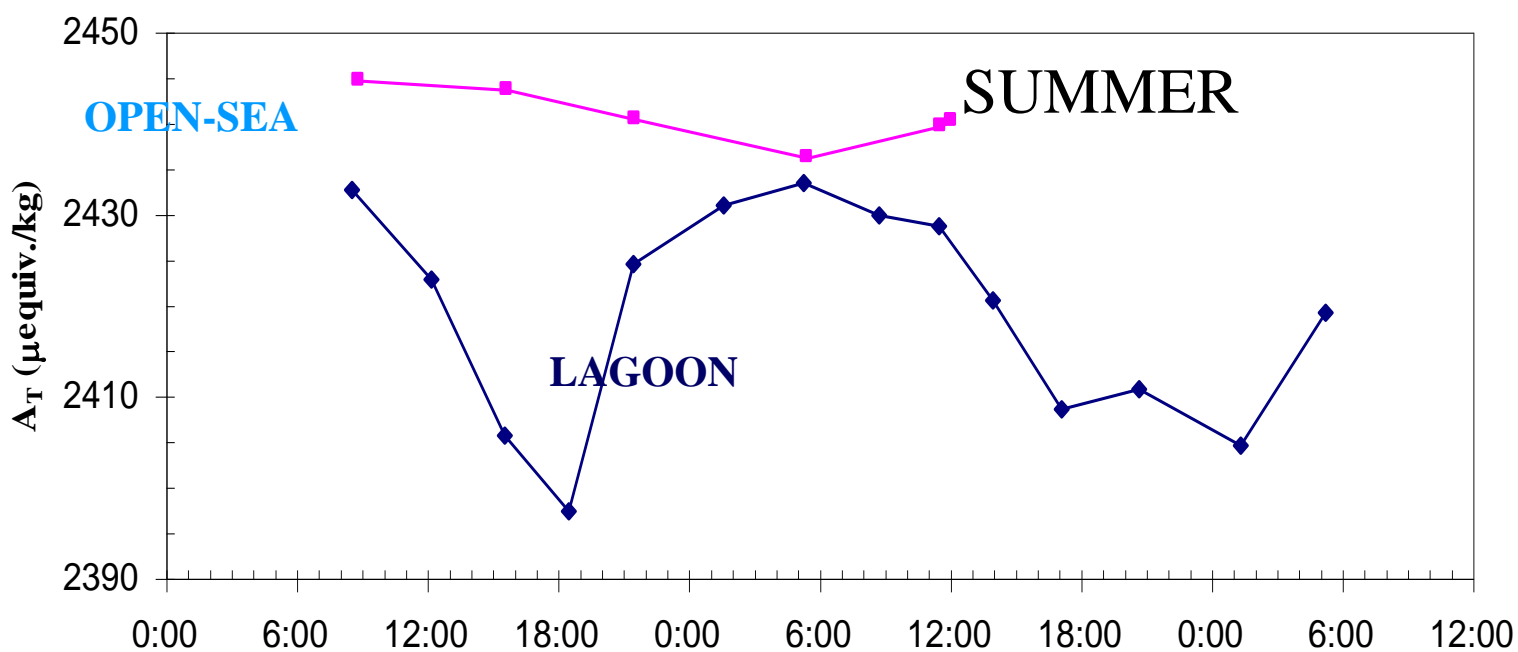


Photosynthesis and respiration

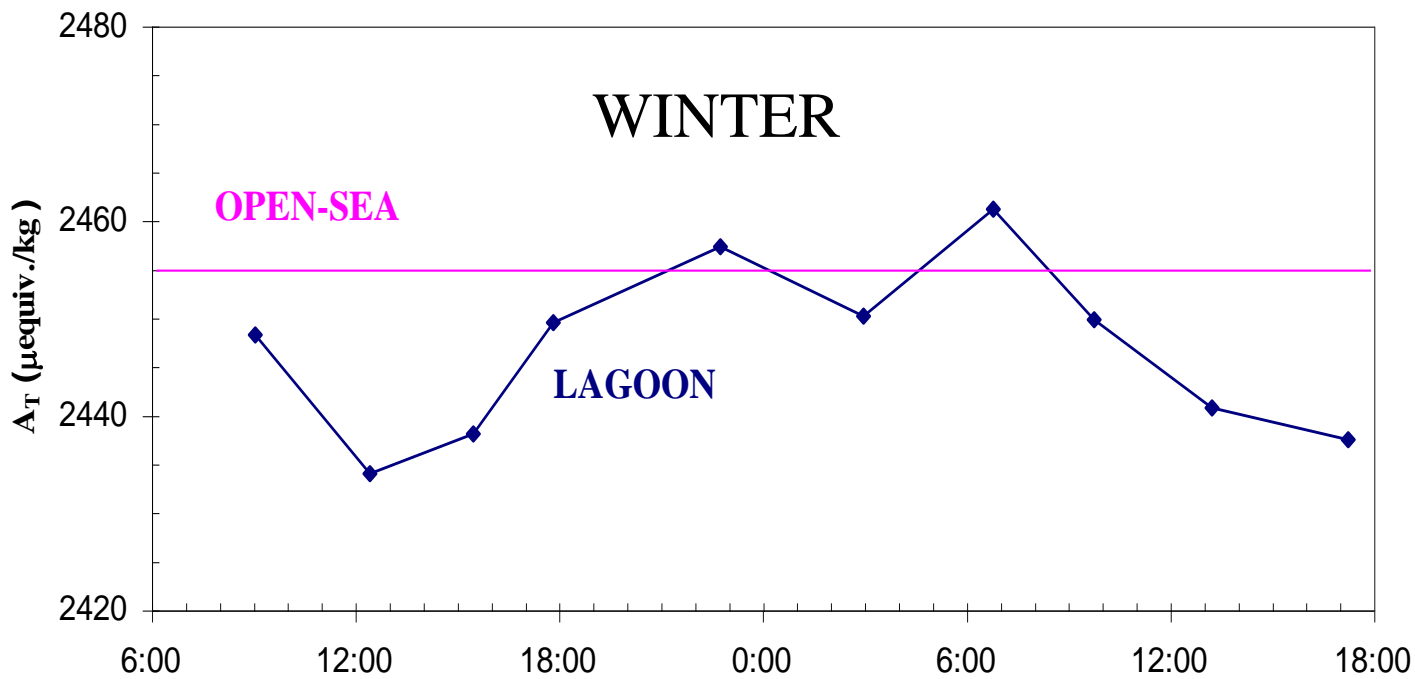
Winter

$P/R \approx 1.5$

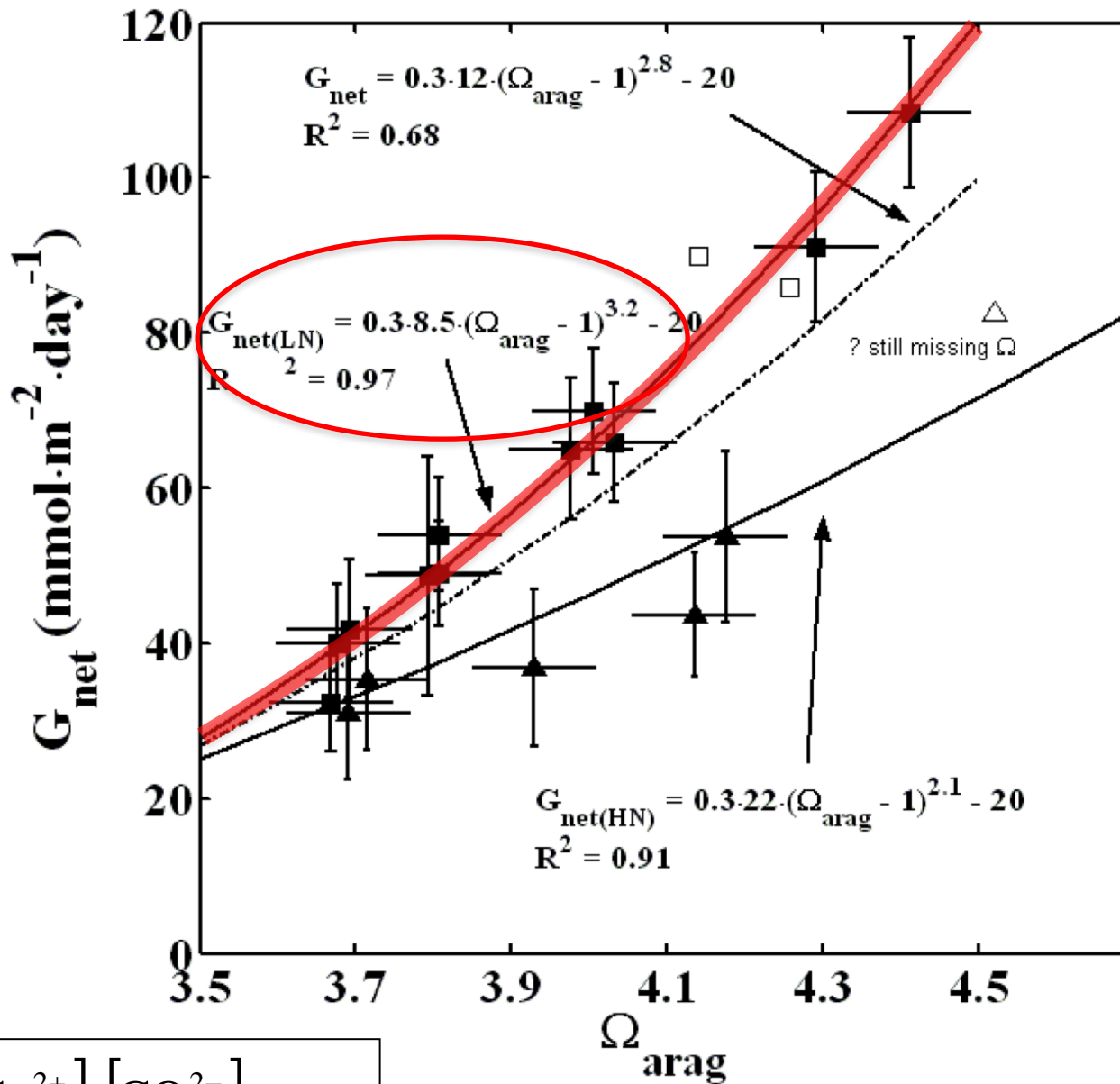




Calcification



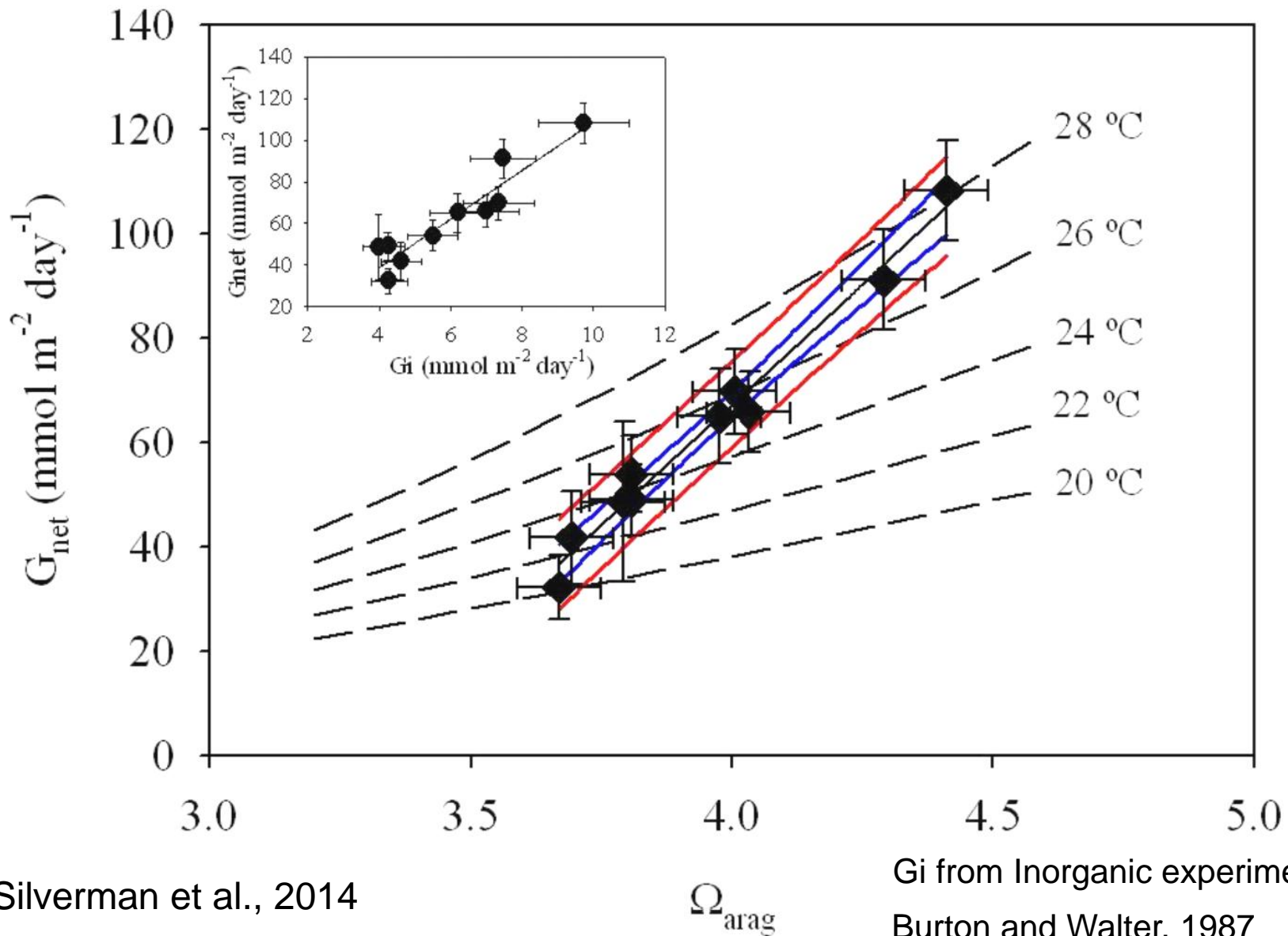
Eilat reef calcification



Jacob Silverman
PhD Thesis

$$\Omega_{\text{arag}} = \frac{[\text{Ca}^{2+}] \cdot [\text{CO}_3^{2-}]}{K_{\text{arag}}} \quad (1)$$

Eilat reef calcification as a function of Ω



Silverman et al., 2014

Ω_{arag}

Gi from Inorganic experiments
Burton and Walter, 1987

The global picture



Photo: I. Greenberg

Oct 27, 2009 9:52 pm

Oct 27, 2009

Nov 24, 2009

Lizard Island

One Tree Island

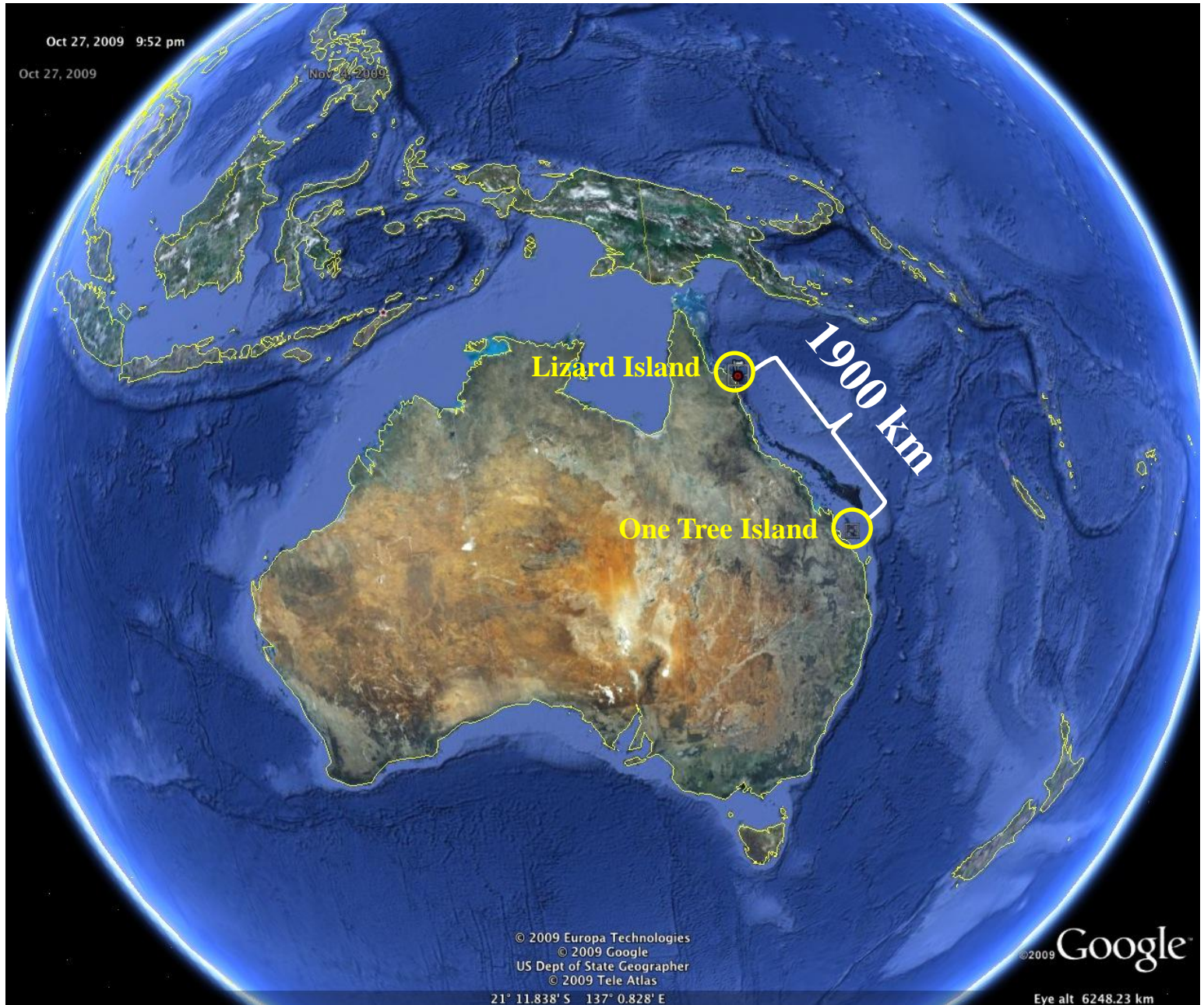
1900 km

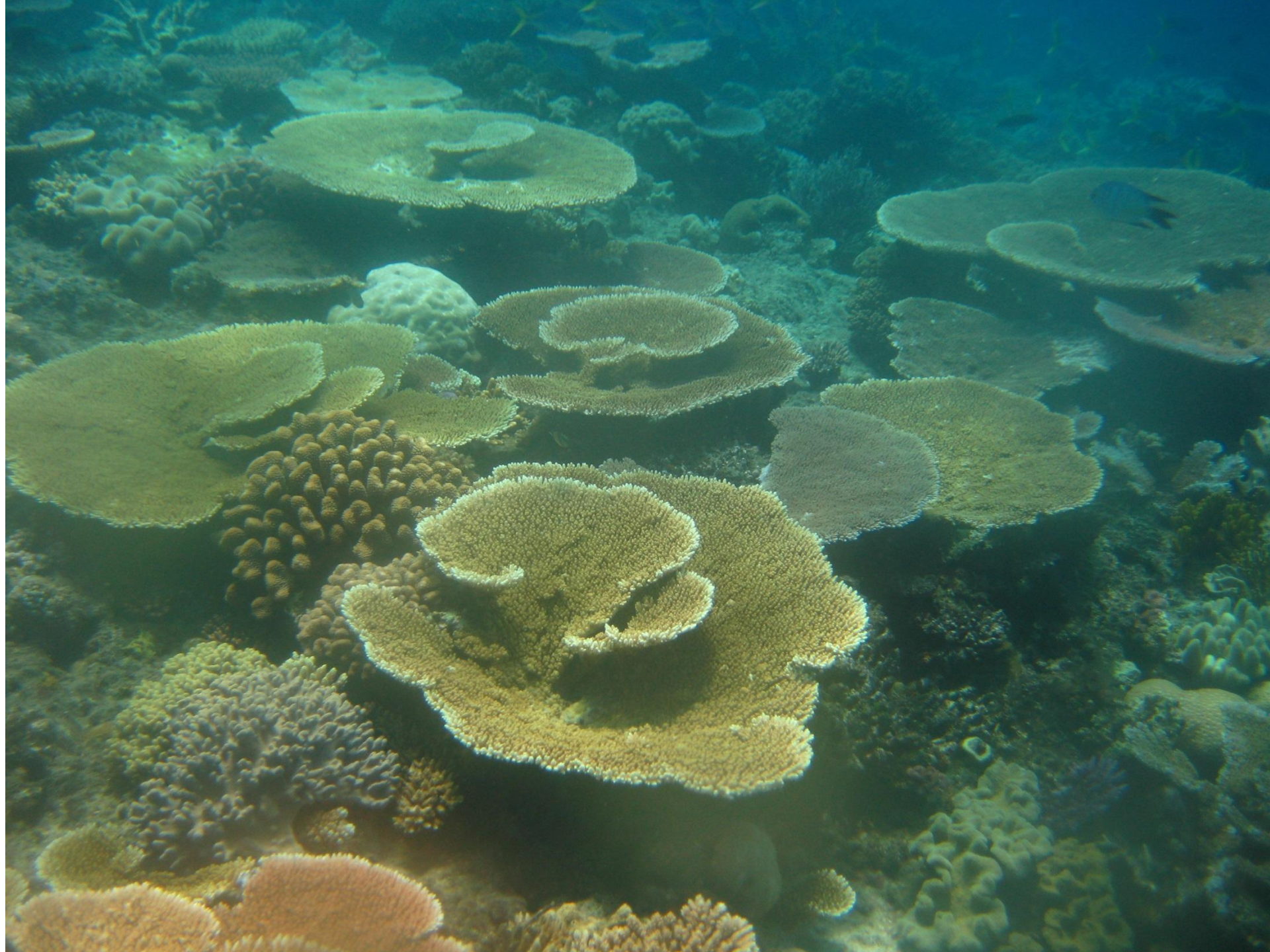
© 2009 Europa Technologies
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US Dept of State Geographer
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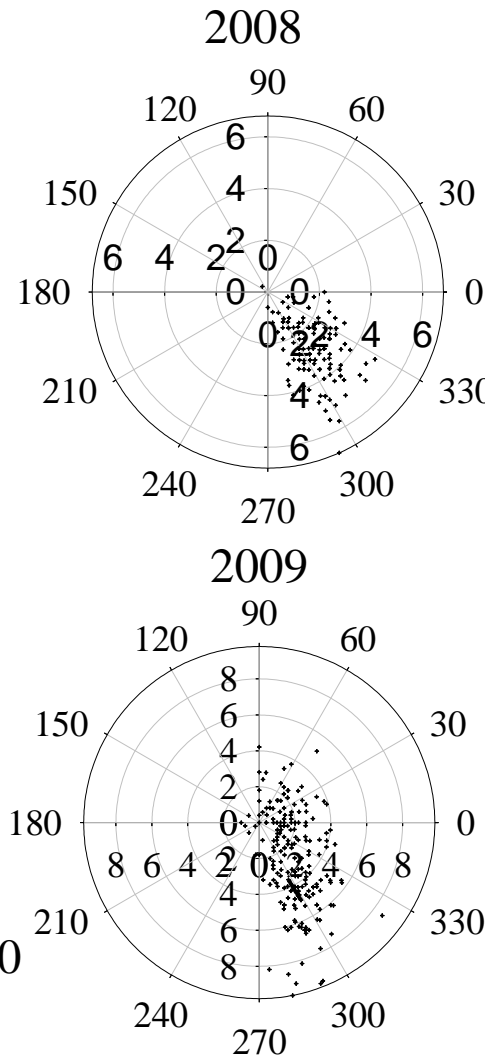
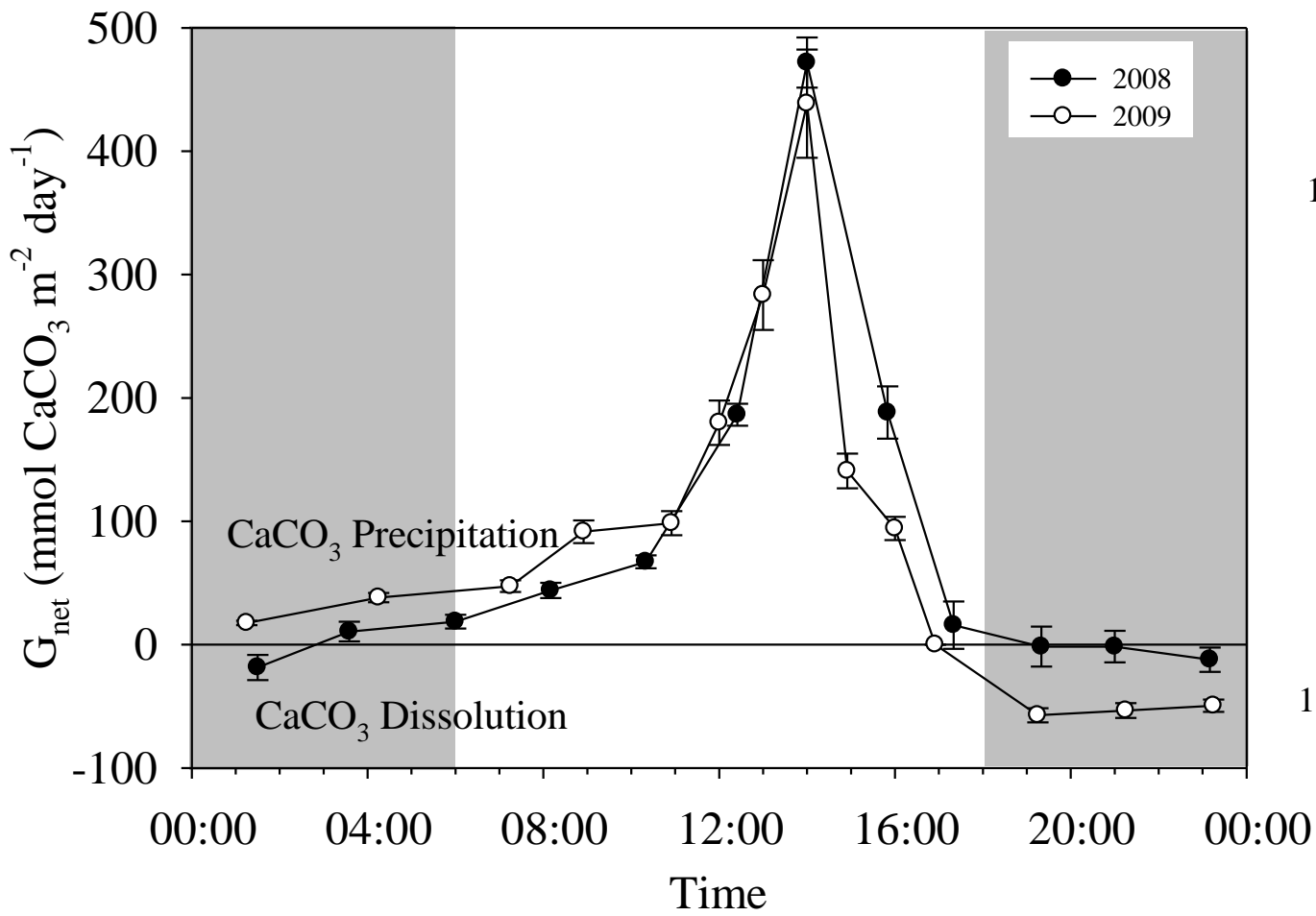
21° 11.838' S 137° 0.828' E

©2009 Google

Eye alt 6248.23 km

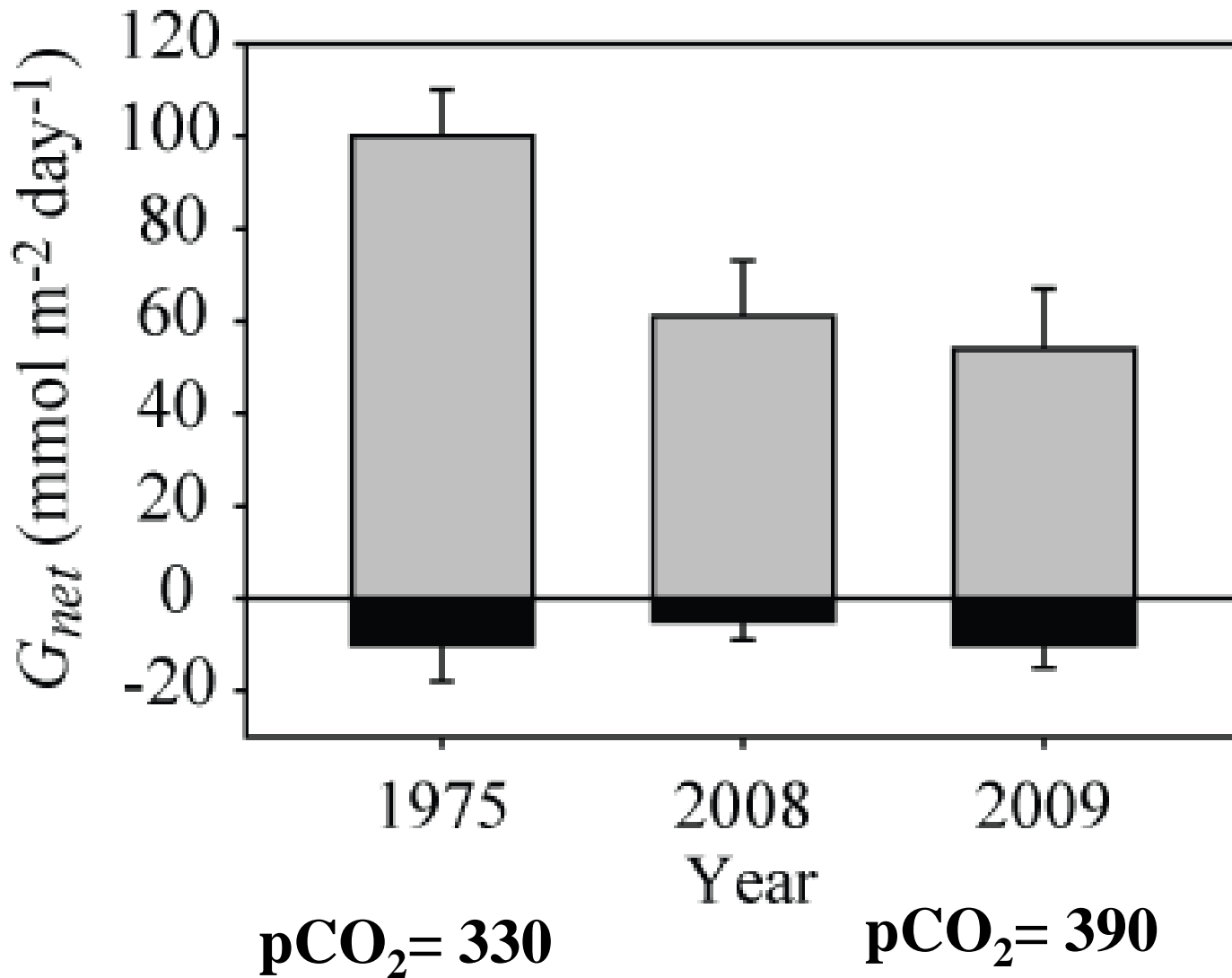


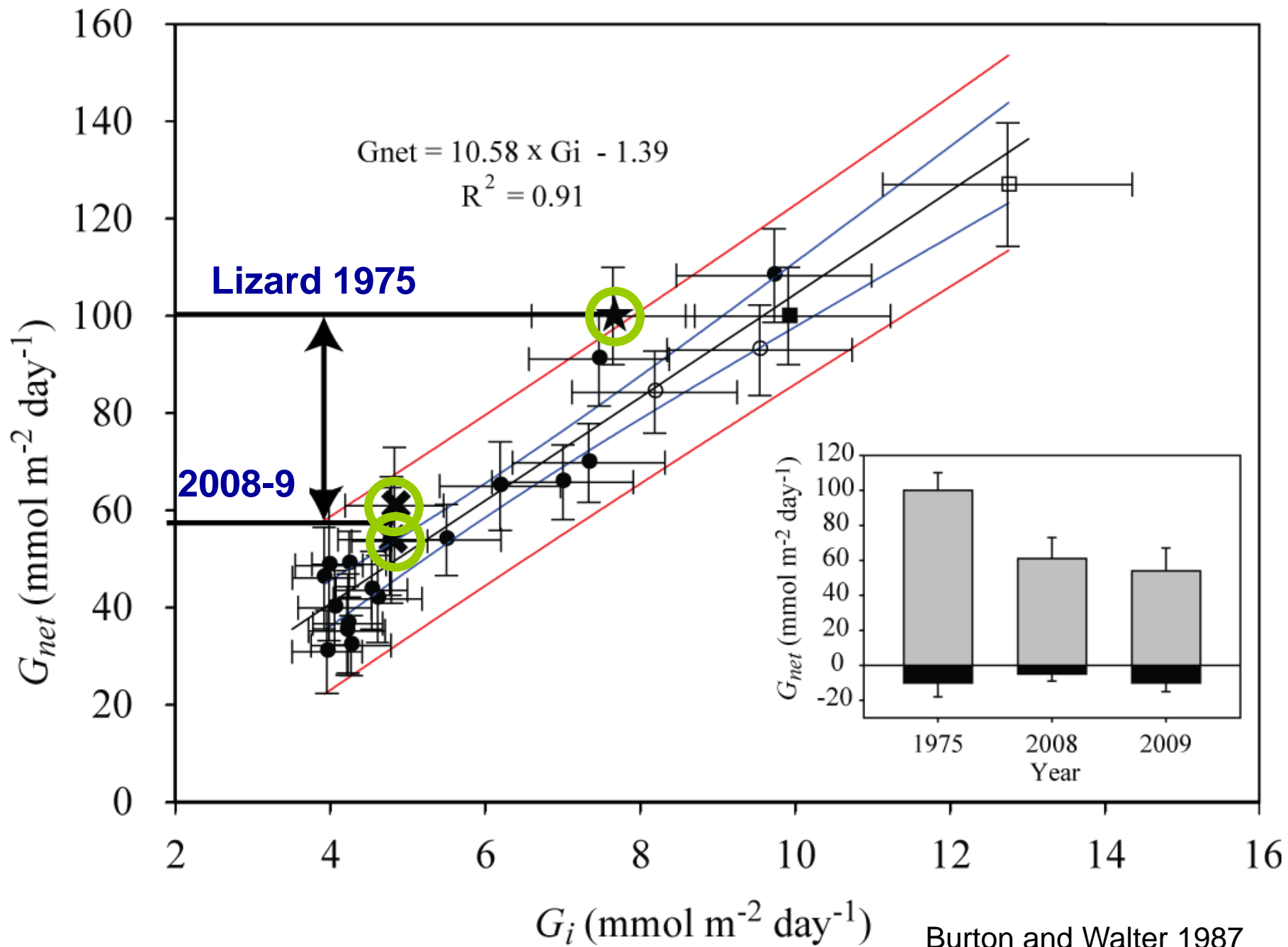




GBR expedition (HUJI –Stanford)

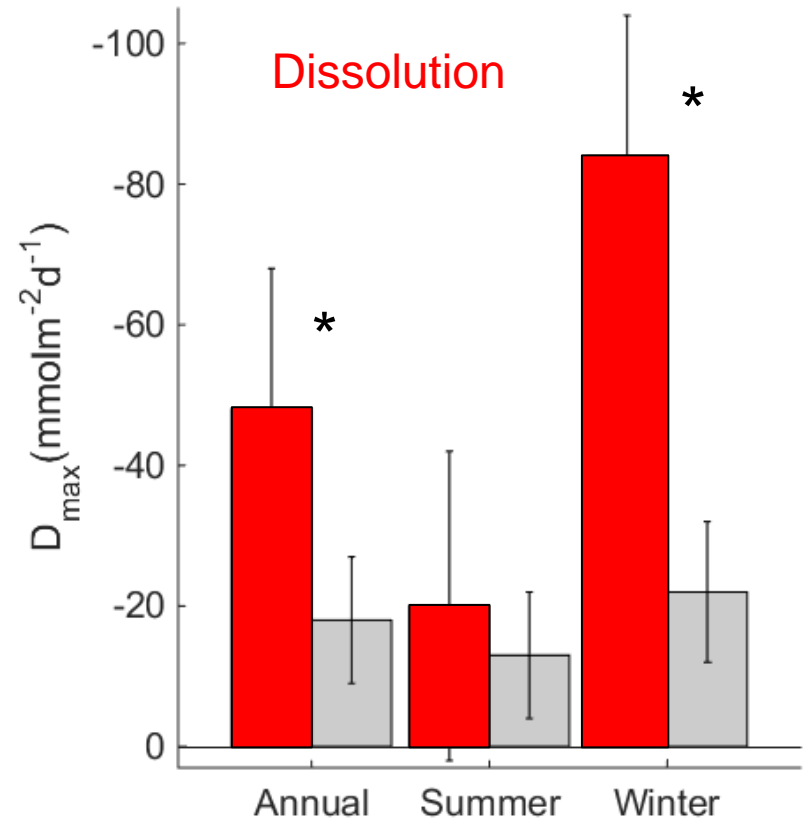
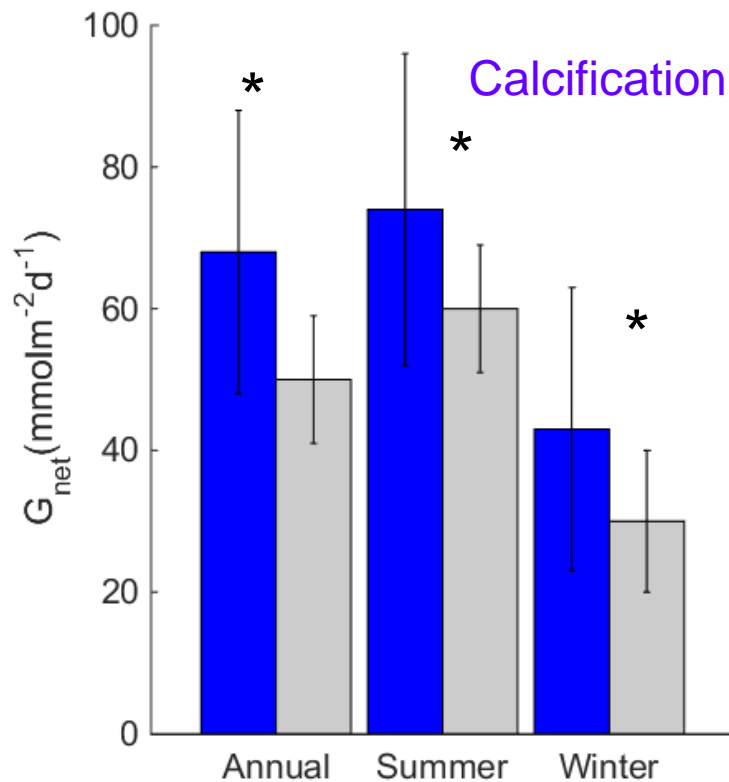
Intensive work in the 1970s by Barnes and Kinsey





Eilat: 15 years perspective

Calcification and Dissolution 2015-6 vs. 2000-2



10-15 % increase in live cover

* $p < 0.05$

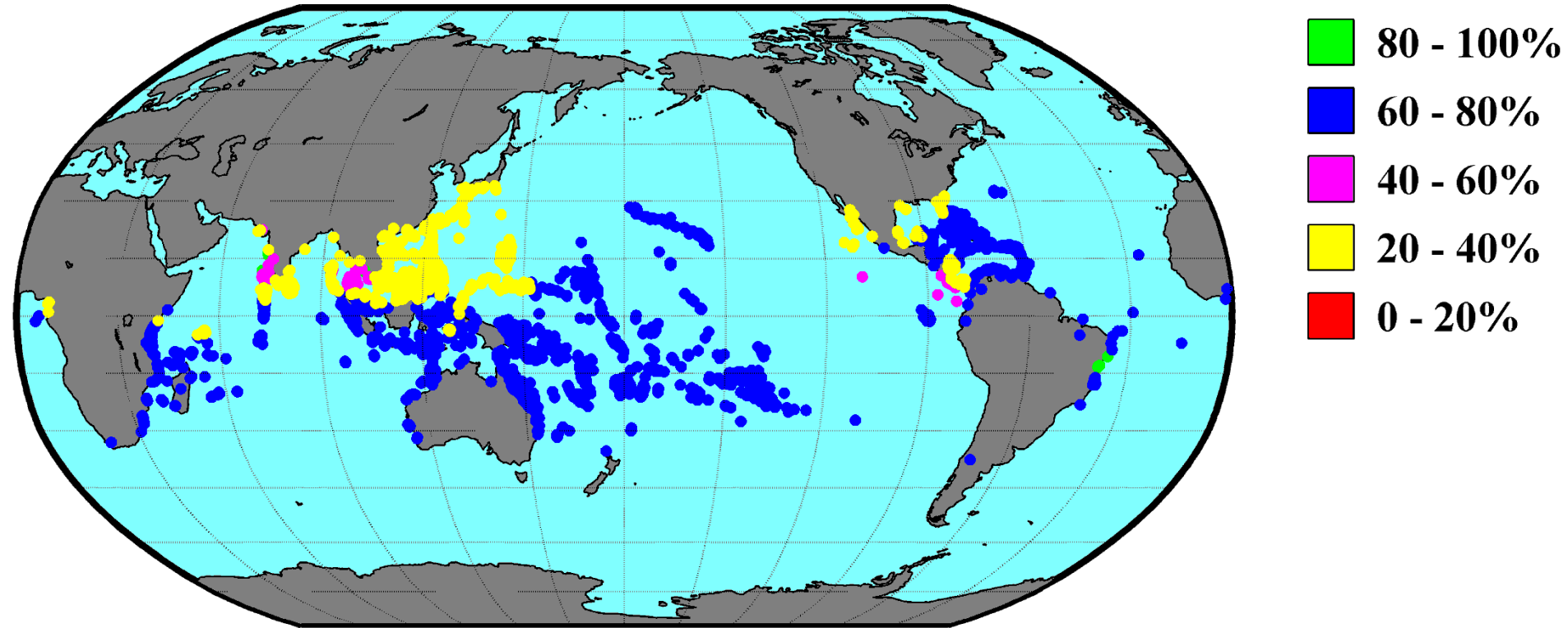
38 ppm increase in atmospheric CO_2



**#OceanAcidification is the
osteoporosis
of the sea.**

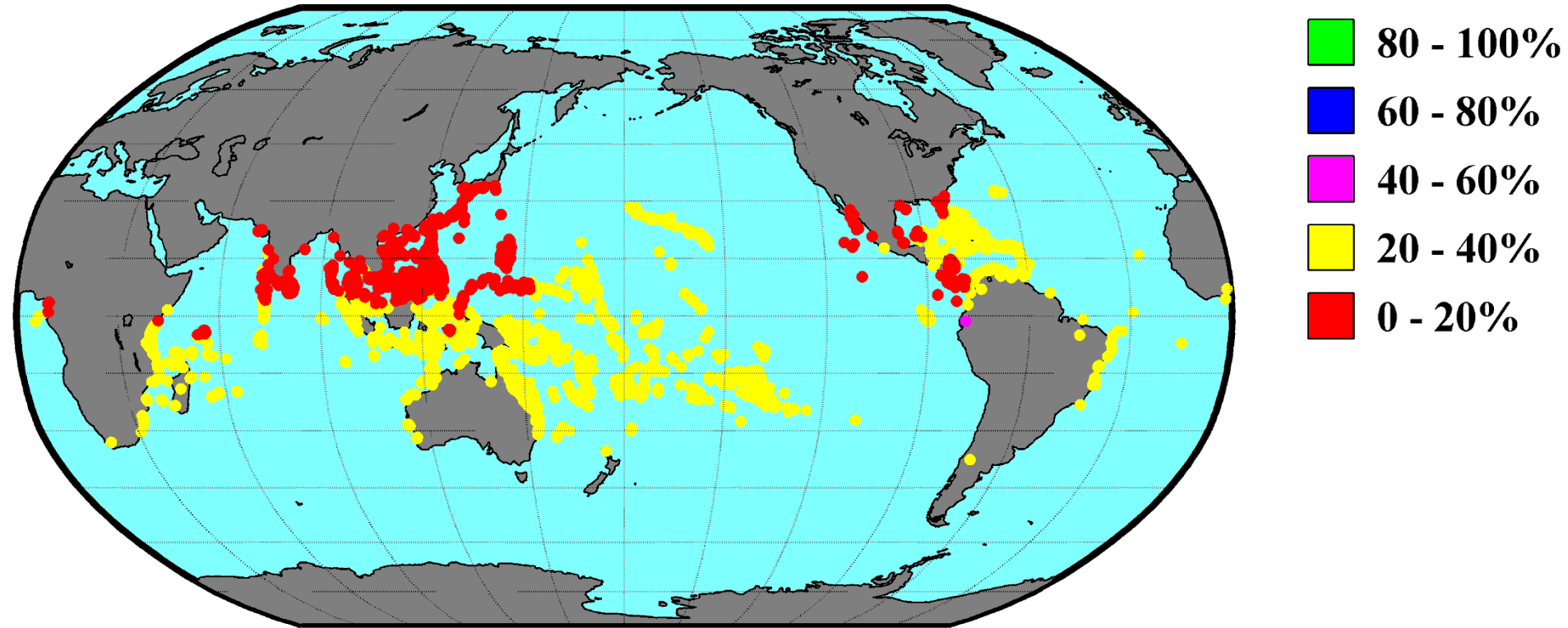
**- Jane Lubchenco
Administrator, National Oceanic and Atmospheric Administration**

$p\text{CO}_2 = 380 \text{ ppm}$, A_c modified by bleaching



Calculated from the Eilat reef data relative the
preindustrial conditions
(280 ppm)

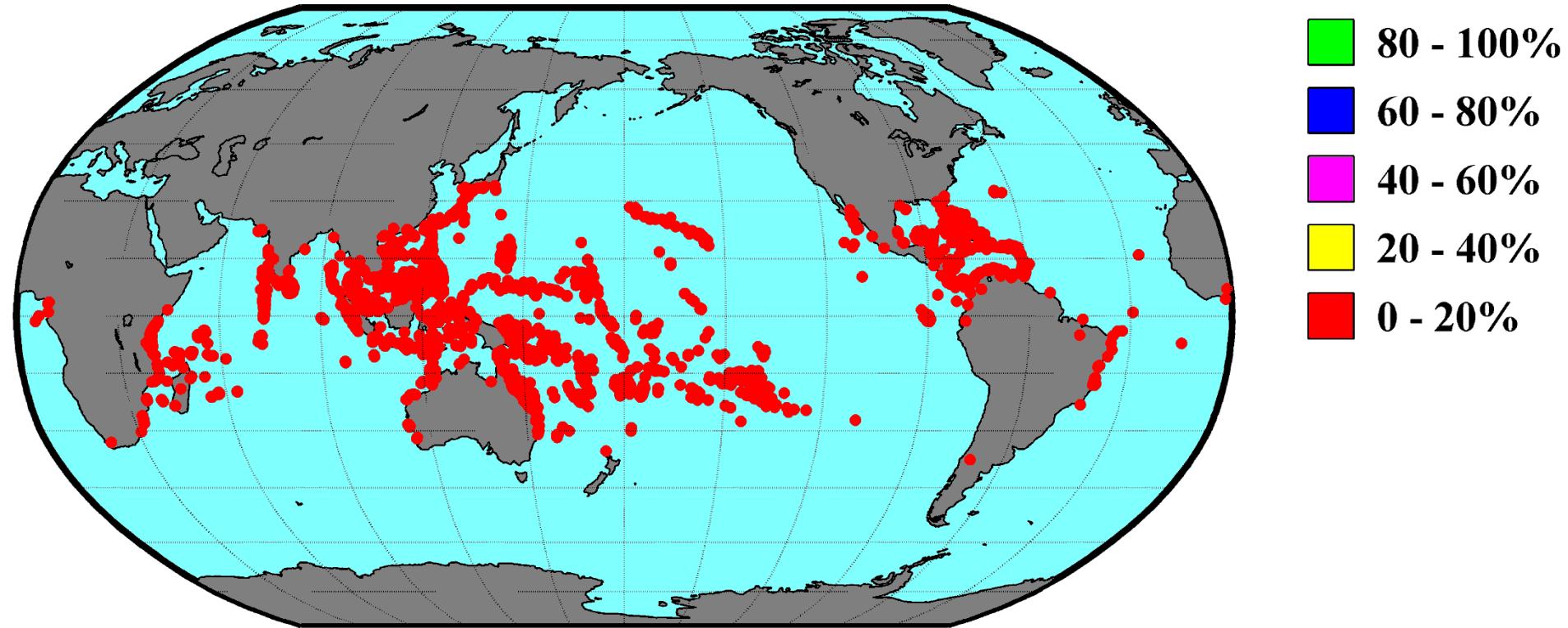
$p\text{CO}_2 = 450 \text{ ppm}$, A_c modified by bleaching



Calculated from the Eilat reef data relative the
preindustrial conditions
(280 ppm)

Silverman et al., 2009 GRL

$p\text{CO}_2 = 560 \text{ ppm}$, A_c modified by bleaching



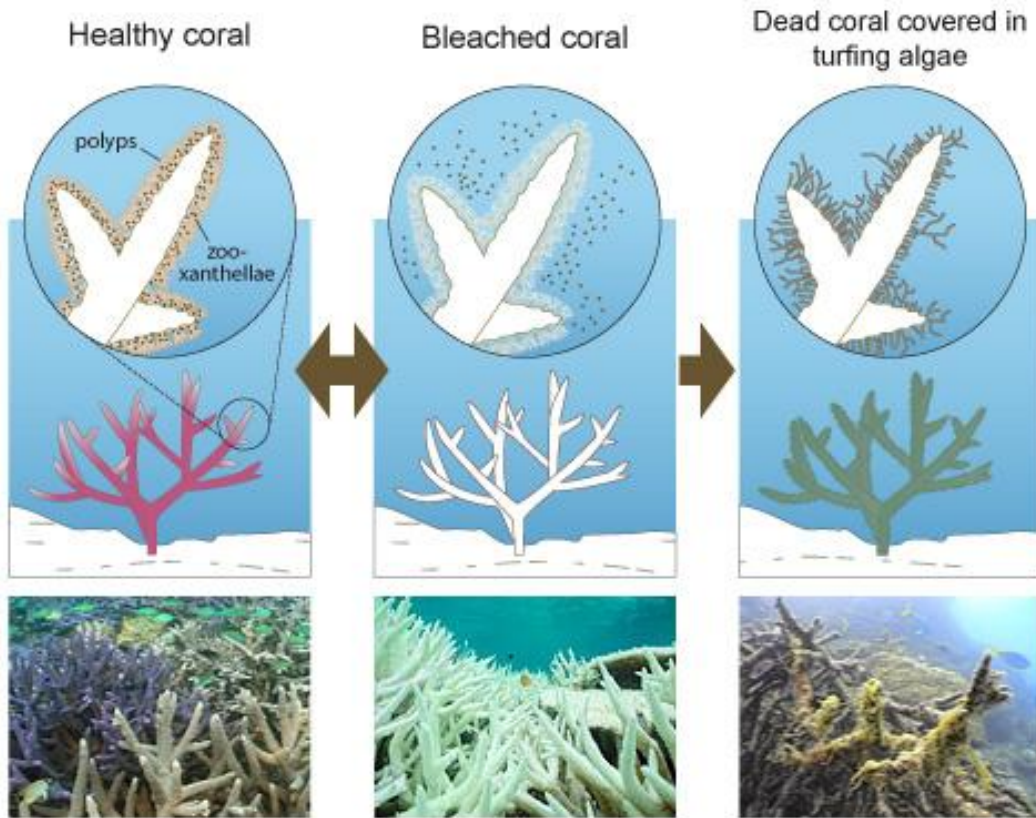
Calculated from the Eilat reef data relative the
preindustrial conditions

(280 ppm)

Silverman et al., 2009 GRL

Coral bleaching

- When temperatures go up ~ 1-2 °C above the summer maximum for a more than a week coral bleach
- The symbionts are expelled from the coral and it loses its pigments becoming bright white
- In most cases the corals die and small reef are turned into rubble
- Bleaching started in 1982 and now it occurs almost every year
- More than 50% of the world corals have died from bleaching already

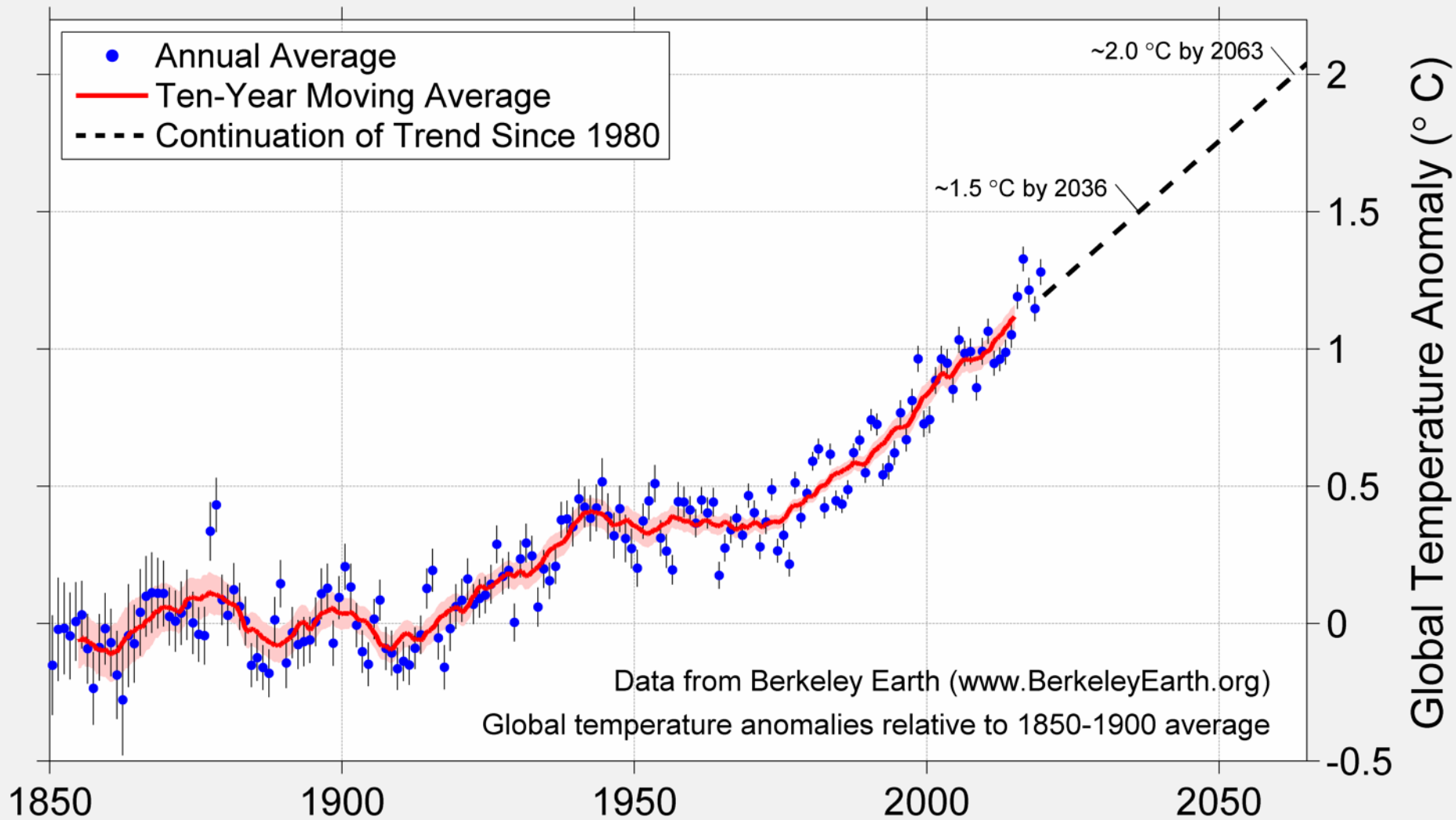


Coral bleaching

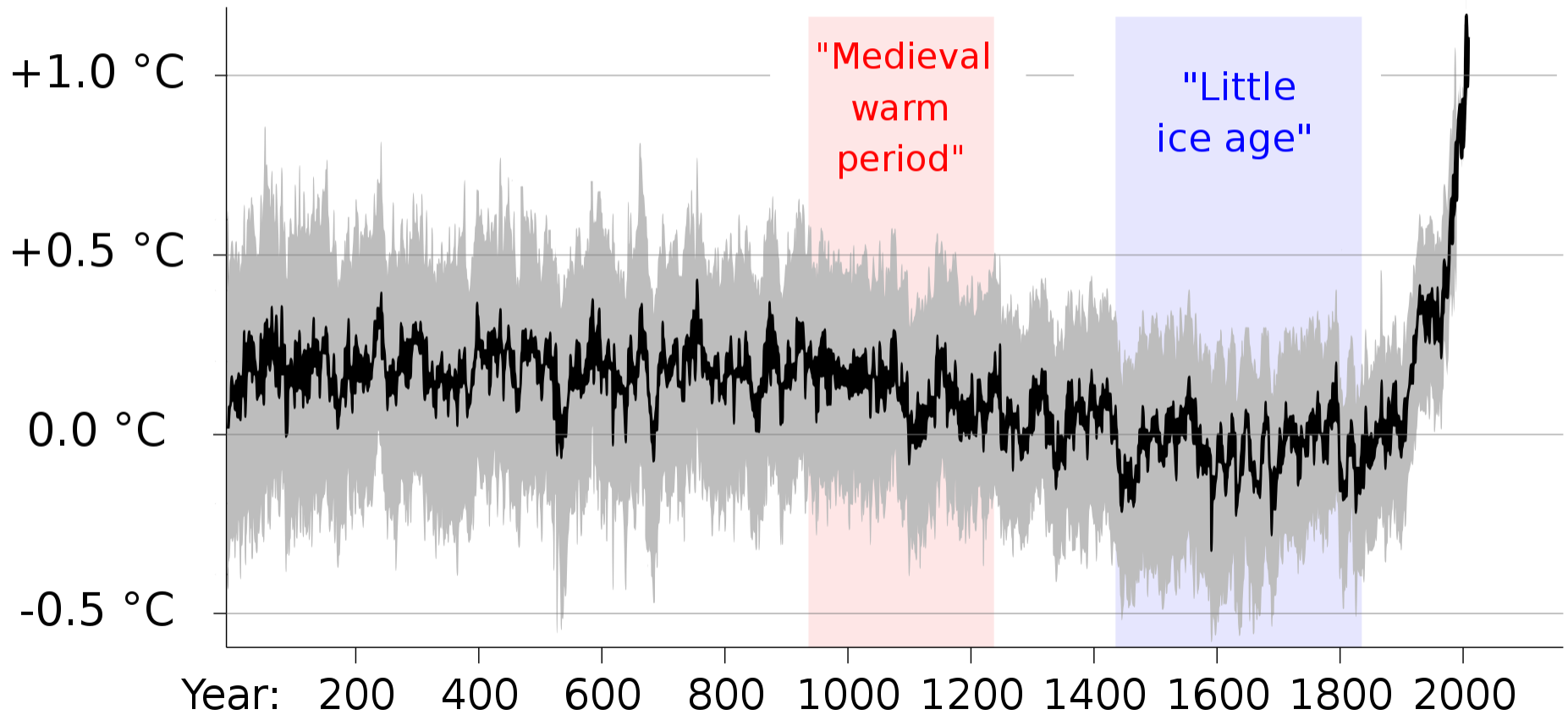


The exact physiological mechanism behind coral bleaching is under debate.

Several mechanisms were proposed: ROS, Apoptosis, membrane fluidity, microbial interactions and more..



Global Average Temperature Change



Why should we care?

Coral bleaching is a highly visual indicator of ocean warming. Most of the extra heat generated by climate change (93%) has been absorbed by the ocean, causing significant shifts in ocean temperature.

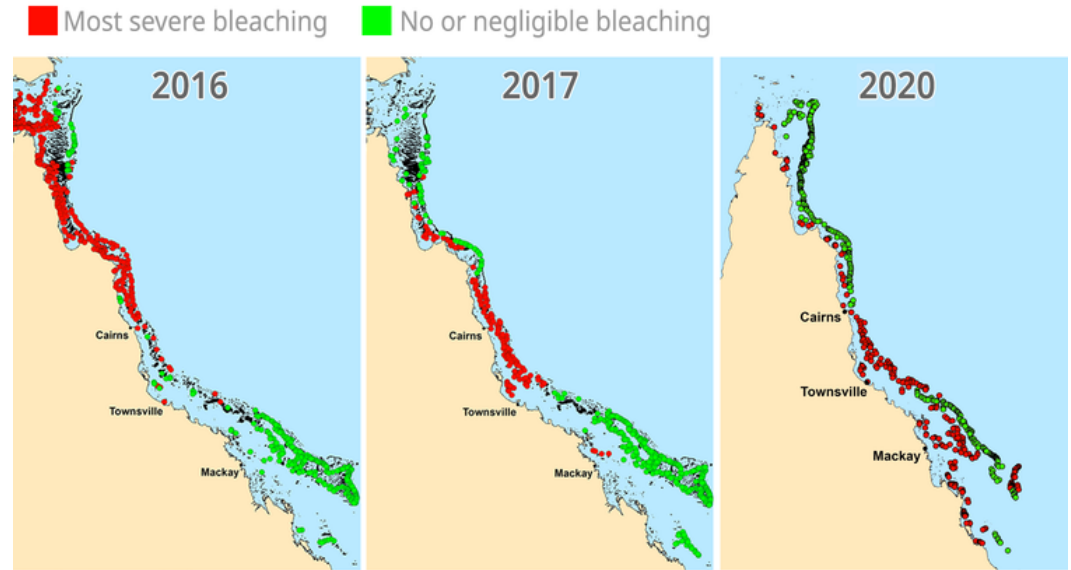
The resulting changes in ocean temperature will increasingly impact weather and climate for decades to come.



XL Catlan Seaview Survey

The last three mass bleaching events

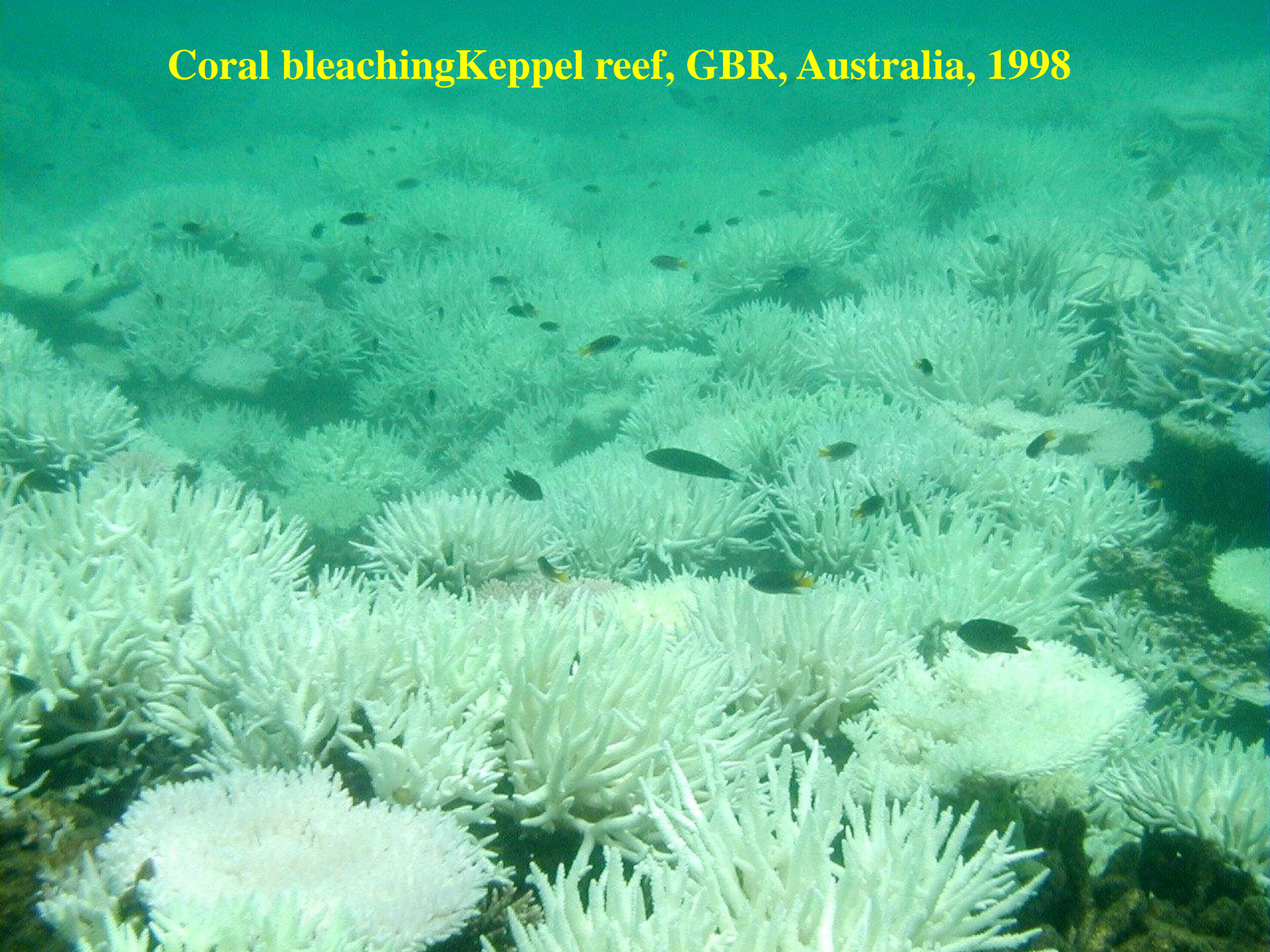
The severity of the last three mass bleaching events on the Great Barrier Reef



Source: ARC Centre of Excellence for Coral Reef Studies



Coral bleaching Keppel reef, GBR, Australia, 1998





Lizard Island soon after bleaching

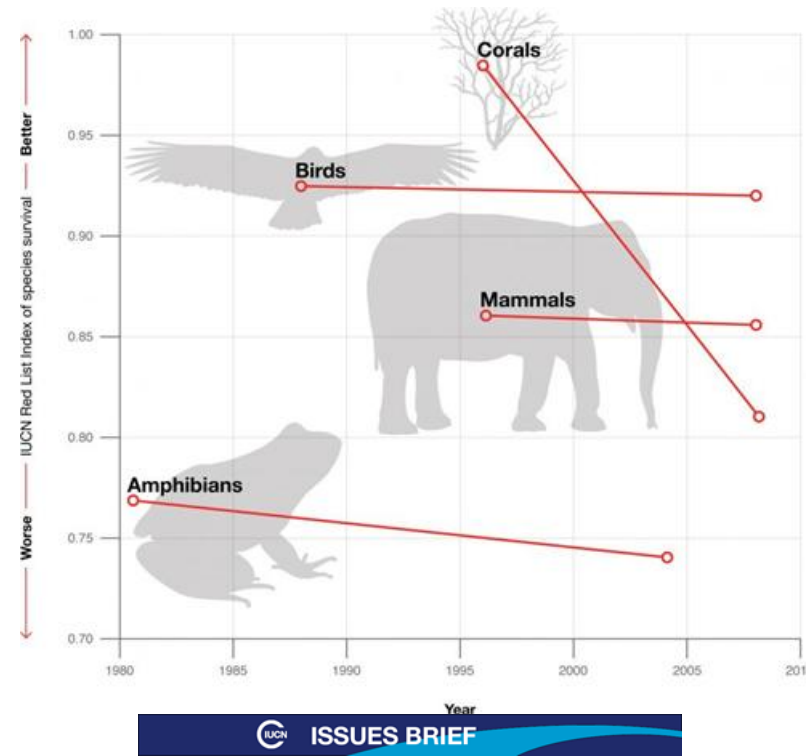


Lizard Island 3 years after bleaching



Summary

- Coral reefs are the most beautiful, diverse and productive ecosystem in the ocean
- Highly productive reefs thrive in the oligotrophic open ocean by extracting plankton as a source of nutrients
- Global changes: eutrophication, ocean acidification and global warming destroy these ecosystems
- More than 50% of the corals in the world have bleached and the rate of bleaching is increasing in time and space
- Coral reefs are perhaps the first marine ecosystems that will disappear due to global change





One Tree Island GBR



One Tree Island, GBR 2009



Lizard Island Expeditions: Sep.-Oct, 2008; Oct.-Nov., 2009

The End